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(54) **Aromatic diamine compounds, bismaleimide compounds, thermosetting resins forming compositions therefrom, resins therefrom, and methods for their preparation.**

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## Description

The present invention relates to aromatic diamine compounds which can be used as raw materials for polyimide, polyamide, bismaleimide or epoxy resins. The compounds can also be employed as curing agents. The invention also relates to a method for preparing the compounds.

Furthermore, the present invention relates to bismaleimide compounds which are useful as raw materials for heat-resistant resins, and which can be obtained from the aforementioned aromatic diamine compounds. The invention also relates to a method for preparing the compounds.

In addition, the present invention relates to thermosetting resin forming compositions, and ultimately the resins, which are obtained from the aforementioned aromatic diamine compounds and bismaleimide compounds, as well as a method for preparing the composition.

In recent years, raw materials for heat-resistant resins have been required to provide a combination of good thermal and mechanical properties, as well as certain other characteristics which composites should exhibit such as flexibility and moldability/workability.

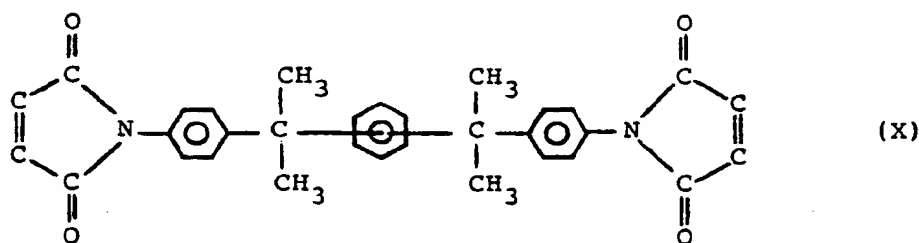
In this regard, polyimide resins have been used. While providing some excellent properties, they typically exhibit poor moldability/workability.

An example of such a polyimide is an aromatic polyimide made by Du Pont and marketed under the trade-name "Vespe". This polyimide can be prepared from 4,4'-diaminodiphenyl ether and pyromellitic anhydride. It is insoluble and unmeltable. Thus, when it is molded, a special procedure such as powder sinter molding must be used. Unfortunately, this molding technique cannot be easily used for the preparation of articles having complex shapes. During manufacturing of complex shaped articles, additional operations such as cutting are necessary. As a result the polyimide is difficult to mold and costs increase.

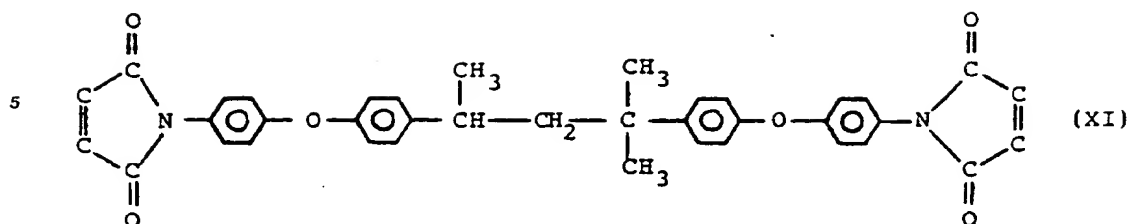
To address those drawbacks associated with polyimide resins, a variety of strategies have been formulated, primarily focusing on improving the diamine component of the raw materials. For example, there have been attempts to introduce an ether linkage group or an isopropylidene group into the molecule and to increase the molecular chain. Unfortunately, these techniques have not effectively provided the flexibility and moldability/workability characteristics necessary for composite materials.

A typical example of a bismaleimide is N,N'-(methylene-di-p-phenylene)bismaleimide disclosed in Japanese Patent Laid-open Nos. 47-8644 and 47-11500. However, this bismaleimide compound is substantially insoluble in common organic solvents such as ketone and petroleum solvents. And, when a polyimide resin is prepared using this compound, the occupation ratio of the polyimido group in the polymer structure is high. Therefore, the polymer is hard, brittle and provides extremely poor flexibility and high hygroscopicity. The effective application of the bismaleimide compounds is accordingly limited considerably.

Recently, in an effort to eliminate such disadvantages, much attention has been given to the development of long-chain bismaleimide compounds. For example, Japanese Patent Laid-open No. 63-500866 discloses a bismaleimide compound having three benzene rings and having the following structure represented by formula (X):



In addition, Japanese Patent Laid-open No. 63-264566 discloses a bismaleimide compound having four benzene rings and having the following structure represented by the formula (XI):



However, even with these bismaleimide compounds and polymers prepared from the compounds, an effective balanced combination of heat resistance, flexibility, adhesion to a metal and inorganic materials, and workability is not provided.

Thermosetting resins having an imido structure have been effectively used to prepare molded articles having excellent electrical insulating properties, heat resistance and dimensional stability. Therefore, this type of resin has been utilized in many industrial fields.

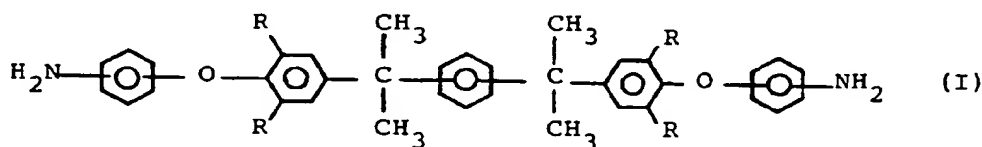
However, thermosetting resins obtained by subjecting aromatic bismaleimides alone to heat polymerization are very brittle and provide poor flexibility, although they do provide excellent heat resistance. To address this drawback, attempts have been made to develop thermosetting resin forming compositions comprising an aromatic bismaleimide and an aromatic diamine. For example, a polyaminobismaleimide resin made by Rhone Poulenc, and marketed under the tradename "Kelimide" comprising N,N'-4,4'-diphenylmethane-bismaleimide and 4,4'-diaminodiphenylmethane has been widely utilized for impregnating varnishes, laminates and molded articles (See Japanese Patent Publication No. 46-23250). However, this type of thermosetting resin is still unsatisfactory in terms of impact resistance and flexibility. Furthermore, when these thermosetting resins are used as base materials for electrical and electronic parts, they exhibit poor moldability/workability and high hygroscopicity, which adversely affect electrical properties.

Therefore, it would be desirable to provide a diamine compound which is useful as a raw material for various resins and which can provide excellent flexibility and moldability/workability. It is likewise desirable to provide a method for preparing such a diamine compound.

It would further be desirable to provide a bismaleimide compound which is readily soluble in common solvents and can be used to prepare a polymer having improved properties. It is likewise desirable to provide a method for preparing such a bismaleimide compound.

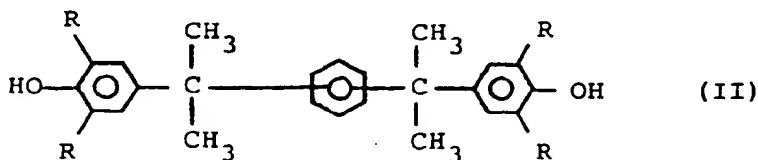
It is still further desirable to provide a thermosetting resin forming composition, as well as a resin, which can provide excellent mechanical strength, particularly in terms of impact resistance, and which has low hygroscopicity. It is also desirable to provide a method for preparing such a composition and resin.

The present invention may enable one to satisfy one or more of these desiderata. The present invention is directed to a novel aromatic diamine compound represented by formula (I):

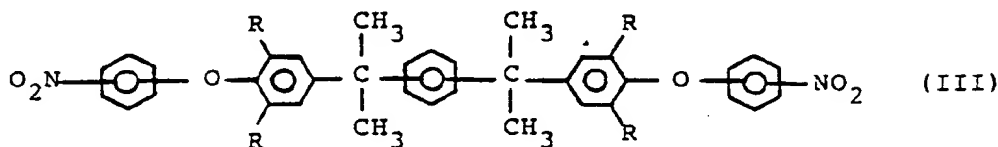


wherein R is H or CH<sub>3</sub>.

The diamine compound of the invention can be obtained by reacting a compound represented by the formula (II):



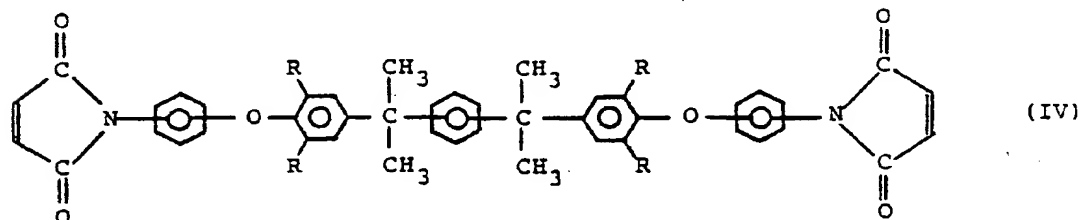
with a nitrobenzene compound such as nitrochlorobenzene or dinitrobenzene in the presence of a base in an aprotic polar solvent to obtain a compound having the formula (III):



The compound of formula (III) is then reduced to obtain the diamine compound of the invention.

The present invention provides a relatively easy technique for providing the novel diamine, the diamine being useful as a raw material for resins, when it may provide excellent flexibility and moldability/workability; or as a curing agent.

Furthermore, the present invention provides a novel bismaleimide compound which comprises a skeleton having five benzene rings and is represented by the formula (IV):



wherein R is H or CH<sub>3</sub>.

The bismaleimide compound represented by the formula (IV) is advantageously soluble in common solvents such as toluene or xylene. A polyimide prepared using this type of bismaleimide may have excellent heat resistance and flexibility, as well as low hygroscopicity.

The bismaleimide of the present invention can be prepared by a dehydration/condensation reaction between a diamine component and maleic anhydride. This dehydration/condensation reaction can be carried out by dissolving or dispersing the diamine of the invention, maleic anhydride and an acid catalyst in a predetermined solvent, and then heating the resulting solution or dispersion under reflux. More specifically, a diamine compound of formula (I) can be reacted with maleic anhydride in the presence of an acid catalyst under heating/reflux of an organic solvent, while any water which is produced is removed from the system.

The bismaleimide compound of the present invention can be used as a raw material to prepare the novel addition-type polyimide of the invention. Since the occupation ratio of the imido group in the molecule of the bismaleimide compound is low, the polyimide resin made from this compound can provide improved flexibility and low hygroscopicity. Therefore, when using the bismaleimide compound of the present invention, blending properties and performance of the thermosetting-type polyimide resin can be considerably improved.

Moreover, by the present invention there is also provided a thermosetting resin forming composition comprising the diamine compound of formula (I) and the bismaleimide compound of formula (IV). The thermosetting resin forming composition of the present invention gives a thermosetting resin which has excellent heat resistance, impact resistance and toughness, as well as low hygroscopicity. Thus, it can be expected that the composition will be useful in the production of electrical and electronic parts, various structural parts, sliding parts and for many other uses. The composition is industrially useful and effective.

#### BRIEF DESCRIPTION OF THE DRAWINGS

Fig. 1 is a mass spectrum of the bismaleimide compound of Example 2A;

Fig. 2 is an IR spectrum of the bismaleimide compound of Fig. 1; and

Figs. 3 and 4 are IR spectra of the bismaleimide compounds of Examples 2C and 2D, respectively.

## DETAILED DESCRIPTION OF THE PREFERRED EMBODIMENTS OF THE INVENTION

According to the present invention, preferable diamine compounds are those compounds of formula (I) where in the amino groups of the aminophenoxy groups are bonded to the meta- or the para positions thereof, and those compounds of formula (I) wherein the two aminophenoxy-(methyl or dimethyl)- $\alpha,\alpha$ -dimethylbenzyl groups are bonded to the meta- or the para positions of the central benzene ring.

Examples of preferable diamine compounds include; (i) 1,4-bis[4-(4-aminophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene, (ii) 1,3-bis[4-(4-aminophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene, (iii) 1,4-bis[4-(3-aminophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene, (iv) 1,3-bis[4-(3-aminophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene, (v) 1,4-bis[4-(4-aminophenoxy)-3,5-dimethyl- $\alpha,\alpha$ -dimethylbenzyl]benzene, (vi) 1,3-bis[4-(4-aminophenoxy)-3,5-dimethyl- $\alpha,\alpha$ -dimethylbenzyl]benzene, (vii) 1,4-bis[4-(3-aminophenoxy)-3,5-dimethyl- $\alpha,\alpha$ -dimethylbenzyl]benzene and (viii) 1,3-bis[4-(3-aminophenoxy)-3,5-dimethyl- $\alpha,\alpha$ -dimethylbenzyl]benzene (hereafter for convenience, "1,4- or 1,3-" will be referred to as "1,4 (1,3)-").

Typically, it has been found that compounds (iii)-(viii) will provide even better flexibility and moldability/workability than the first two compounds (i) and (ii).

There are known techniques for reducing a bis(4-nitrophenoxy) compound obtained by reacting a bisphenol with 4-chloronitrobenzene in order to prepare a corresponding diamine. For example, bis(4-aminophenoxy)biphenyl has been manufactured by reducing bis(4-nitrophenoxy)biphenyl obtained from 4,4'-biphenol and 4-chloronitrobenzene [28th National Sampe Symposium, April 12-14, p. 728-736 (1983)]. Furthermore, 2,2'-bis(4-aminophenoxy)propane has been manufactured from 2,2'-bis(4-hydroxyphenyl)propane [Roxs Chem., 48, p. 1459 (1974)]. However, a diamine compound having a structure of formula (I) according to the present invention has not been so produced.

In order to obtain a diamine compound of formula (I), it is necessary to pass through two steps of condensation and reduction. For example, when producing 1,4 (1,3)-bis[4-(4-aminophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene, 1,4 (1,3)-bis(4-hydroxy- $\alpha,\alpha$ -dimethylbenzyl)benzene, which has a structure represented by formula (II), is reacted with 4-chloronitrobenzene in the presence of a base in an aprotic polar solvent in order to prepare 1,4 (1,3)-bis[4-(4-nitrophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene, which has a structure of formula (III).

The amount of 4-chloronitrobenzene is preferably two to three moles, more preferably 2.05 to 2.5 moles, per mole of 1,4 (1,3)-bis(4-hydroxy- $\alpha,\alpha$ -dimethylbenzyl)benzene.

Examples of suitable bases include carbonates, hydrogen carbonates, hydroxides and alkoxides of alkaline metals. Typical examples of these bases include potassium carbonate, sodium carbonate, potassium hydrogen carbonate, sodium hydrogen carbonate, potassium hydroxide, sodium hydroxide, lithium hydroxide, sodium methoxide and potassium isopropoxide. Among these, potassium carbonate, sodium hydroxide and potassium hydroxide are most preferably used and are also preferred from an economical viewpoint. The amount of the base is preferably not less than 2 equivalents, more preferably 2.2 to 3 equivalents when 1,4 (1,3)-bis(4-hydroxy- $\alpha,\alpha$ -dimethylbenzyl)benzene is the raw material.

As the solvent used in the above described condensation reaction, aprotic polar solvents are suitable. Typical examples of these solvents include dimethyl sulfoxide, N,N-dimethylformamide, N,N'-dimethylacetamide, N-methylpyrrolidone, 1,3-dimethyl-2-imidazolidinone, hexamethylphosphortriamide and sulfolane. The amount of the solvent is preferably from 1 to 15 times, more preferably 2 to 5 times, the total weight of raw materials.

The reaction temperature may range from 50 to 220°C, and is preferably in the range of 80 to 180°C.

The end point of the reaction can be determined by observing a decrease of raw materials and unreacted intermediate through thin-layer chromatography or high-speed liquid chromatography.

This exemplified condensation reaction may be carried out by placing predetermined amounts of 1,4 (1,3)-bis(4-hydroxy- $\alpha,\alpha$ -dimethylbenzyl)benzene, the base and the solvent in a reactor and forming an alkaline metal salt of 1,4 (1,3)-bis(4-hydroxy- $\alpha,\alpha$ -dimethylbenzyl)benzene. Then, 4-chloronitrobenzene can be added thereto. Alternatively, all the raw materials, including 4-chloronitrobenzene, can be initially placed in the reactor. However, the reaction is not limited to these procedures, and can be carried out using other suitable procedures.

After completion of the reaction to obtain the intermediate represented by the formula (III), (e.g., 1,4 (1,3)-bis[4-(4-nitrophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene), optional steps can be performed. For example, the concentration of the solvent can be modified or the reaction solution can be discharged into water.

The intermediate of formula (III) is then reduced to prepare the desired compound represented by the formula (I) (e.g., 1,4 (1,3)-bis[4-(4-aminophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene). The reduction can be achieved by standard procedures such as those which involve reducing a nitro group to an amino group. In an industrial context, catalytic reduction or hydrazine reduction is preferred.

In the case of catalytic reduction, the reduction reaction is carried out using a reduction catalyst such as a metallic catalyst (which may be on a carrier), e.g. nickel, palladium or platinum or a Raney catalyst of nickel

or copper in an amount of 0.01 to 10% (by weight) in terms of the metal with respect to the intermediate, e.g., 1,4(1,3)-bis[4-(4-nitrophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene.

In the reduction reaction a solvent is used. Preferable examples of the solvent include methanol, ethanol, isopropyl alcohol, methyl cellosolve and N,N-dimethylformamide which are inactive in the reaction. There is no particular limitation on the reaction temperature, but it may be in the range of from 20 to 200°C, preferably 20 to 100°C. The reaction pressure may be from atmospheric to about 50 atm. ( $1 \times 10^5$  -  $50 \times 10^5$  Nm<sup>-2</sup>).

In the case of hydrazine reduction, the reduction reaction is carried out using hydrazine in amount which exceeds the theoretical amount by 1.1 to 5 times.

The catalyst can be a metallic catalyst usually employed for catalytic reduction. The particularly preferable catalysts are palladium/carbon, platinum/carbon or that which is prepared by allowing active carbon to adsorb ferric chloride. The amount of the catalyst may range from 0.001 to 30% (by weight) in terms of metal with respect to the intermediate, e.g., 1,4(1,3)-bis[4-(4-nitrophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene.

As a reaction solvent, there can be used in those solvents which are known for use in catalytic reduction. There is no particular restriction on reaction temperature. But it may usually be in the range of from 20 to 150°C, preferably 40 to 120°C.

After completion of the reaction, the reaction solution is hot-filtered and can be subjected to optional treatments such as cooling, concentration, dilution and crystallization in order to obtain desired crystals of the product (e.g., 1,4(1,3)-bis[4-aminophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene). The crystals can be purified by suitable techniques such as washing and recrystallization.

The novel diamine according to the invention (e.g., 1,4(1,3)-bis[4-(4-aminophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene) can be used to prepare polyimide resins, polyamide resins, bismaleimide resins or an epoxy resin using known techniques.

It will be appreciated that 1,4(1,3)-bis[4-(4-aminophenoxy)-3,5-dimethyl- $\alpha,\alpha$ -dimethylbenzyl]benzene, which is one of the diamine compounds of the present invention, can be prepared by the above-mentioned procedure except that 1,4(1,3)-bis(4-hydroxy-3,5-dimethyl- $\alpha,\alpha$ -dimethylbenzyl)benzene is used as the compound of the formula (II). Furthermore, 1,4(1,3)-bis[4-(3-aminophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene can be prepared by the above-mentioned procedure except that 4-chloronitrobenzene is replaced with m-dinitrobenzene. In other words, various diamine compounds of the present invention can be obtained by varying the combination of the compound of formula (II) and the nitrobenzene compound.

The bismaleimide compound of the present invention can be prepared using a dehydration/condensation reaction between a diamine component represented by formula (I) and maleic anhydride. As a result of this reaction, amino groups present at the opposite ends of formula (I) are replaced with maleimide groups.

The dehydration/condensation reaction involves dissolving or dispersing the diamine, maleic anhydride and an acid catalyst in a predetermined solvent. The resulting solution is heated under reflux.

Preferable examples of the acid catalyst include mineral acids such as sulfuric acid, hydrochloric acid and phosphoric acid; heteropolyacids such as phosphotungstic acid and phosphomolybdic acid; organic sulfonic acids such as p-toluenesulfonic acid and methanesulfonic acid; halogenated carboxylic acids such as trichloroacetic acid and trifluoroacetic acid; solid acids such as silica-alumina; and cation exchange type ion exchange resins. In particular, sulfuric acid, phosphoric acid and p-toluenesulfonic acid are preferred. In addition, the acids may be in the form of salts with diamines.

The amount of acid catalyst depends upon its type. But it may usually be in the range of from 0.1 to 10% by weight with respect to the total weight of maleic anhydride and the diamine. When the amount of the catalyst is less than 0.1% by weight, the effect of the catalyst will not always be achieved. When the amount is increased to more than 10% by weight, no additional effect may be obtained from the catalyst, which is uneconomical. In addition, it is inconvenient and difficult to remove the excess of catalyst.

Examples of suitable solvents to be used in the condensation reaction include aliphatic and alicyclic hydrocarbons such as hexane, heptane, decane and cyclohexane; aromatic hydrocarbons such as benzene, toluene and xylene; halides of these aliphatic and aromatic hydrocarbons; oxygen-containing, nitrogen-containing and sulfur-containing polar solvents such as dimethylformamide, N-methylpyrrolidone, acetonitrile, dimethylacetamide, dimethyl sulfoxide, sulfolane, anisole and n-butyl ether; and mixtures thereof. The amount of the solvent is preferably 1 to 20 times, particularly preferably 3 to 10 times as much as the total weight of the diamine component and maleic anhydride.

The reaction temperature under heating and reflux depends slightly upon the type of solvent. However, it may usually be in the range of from 80 to 190°C, preferably from 100 to 160°C. High pressure, atmospheric pressure or a reduced pressure can all be used, and will typically be selected to be compatible with the type of solvent and reaction temperature.

The reaction time may usually be in the range of from 2 to 10 hours, preferably 5 to 6 hours.

With regard to the amounts of maleic anhydride and diamine component, it is desirable that the amount

of maleic anhydride is slightly in excess relative to that of the diamine component. In general, the molar ratio of maleic anhydride to diamine is preferably 2.05 to 3.0.

After completion of the condensation reaction, the reaction mixture is washed with water to remove any remaining catalyst and unreacted maleic anhydride therefrom. The solvent is then distilled off, thereby obtaining a concentrate. Afterward, alcohol is added to the concentrate, followed by stirring, so that a crystal powder is obtained. The resulting crude crystals can then be purified by recrystallization.

In order to obtain the thermosetting resin forming composition, and ultimately the resin, of the present invention from the bismaleimide compound and the diamine compound, it is preferred that a bismaleimide compound represented by formula (IV) be mixed with an aromatic diamine compound represented by formula (I) in the absence of any solvent or in the presence of an organic solvent. The mixture can then be subjected to heat treatment at a temperature of from 70 to 220°C to carry out prepolymerization; the mixture and the prepolymer both being "resin forming compositions". Thereafter, the resin can be produced.

The following procedures are preferably employed. (1) The bismaleimide compound and the aromatic diamine compound can be mixed and then ground. The ground mixture may be used as such. Alternatively, the mixture can be further subjected to heat treatment to form a prepolymer. The prepolymer can then be ground to pellets or powder. In this case, heating conditions wherein the material is partially cured to the level of the prepolymer are preferable. In general, it is appropriate that the heat treatment is effected at a temperature of from 70 to 220°C for 5 to 240 minutes, preferably at 80 to 200°C for 10 to 180 minutes.

Or, (2) The bismaleimide compound and the aromatic diamine compound can be dissolved in an organic solvent. Then, the solution can be discharged into a poor solvent. The resulting precipitate can be collected by filtration, dried and then formed into pellets or powder. Alternatively, the bismaleimide compound and the aromatic diamine compound can be dissolved in an organic solvent, and then partially cured up to the level of a prepolymer by heat treatment. Afterward, the cured material can be discharged into a poor solvent. The resulting precipitate can be collected by filtration, dried and then formed into pellets or powder. Conditions for the heat treatment depend slightly upon the type of organic solvent, but they are generally consistent with those used for heat treatment according to procedure (1).

The organic solvent should not substantially react with both components, but should be a good solvent for both components. Examples of suitable solvents include halogenated hydrocarbons such as methylene chloride, dichloroethane and trichloroethylene; ketones such as acetone, methyl ethyl ketone, cyclo-hexanone and di-isopropyl ketone; ethers such as tetra-hydrofuran, dioxane and methyl cellosolve; aromatic compounds such as benzene, toluene and chlorobenzene; and aprotic polar solvents such as acetonitrile, N,N-dimethylformamide, N,N-dimethylacetamide, dimethyl sulfoxide, N-methyl-2-pyrrolidone and 1,3-dimethyl-2-imidazolidinone.

The ratio between the bismaleimide compound and the diamine compound is from 0.1 to 1.2 moles, preferably 0.2 to 0.8 mole of diamine compound per mole of the bismaleimide compound. When the ratio is too small, the resulting cured product will not have good impact resistance or flexibility. When it is too large, the cured product is adversely affected in terms of heat resistance.

Additional components can be added to the thermosetting resin forming composition of the present invention, so long as the objects of the present invention are not impaired. For example, a curing accelerator can be added. Suitable accelerators include radical polymerization initiators such as azo compounds or organic peroxides, or ionic catalysts such as tertiary amines, quaternary ammonium salts, imidazoles or boron trifluoride-amine salts.

Powdery reinforcing material and fillers can also be used. Examples include metal oxides such as aluminum oxide or magnesium oxide; metal hydroxides such as aluminum hydroxide; metal carbonates such as calcium carbonate or magnesium carbonate; diatomaceous earth powders; basic magnesium silicates; calcined clays; finely powdered silica; melted silica; crystalline silica, carbon black; kaolin; finely powdered mica; quartz powder; graphite; asbestos; molybdenum disulfide; or antimony trioxide.

Also, fibrous reinforcing materials and fillers can be added. Examples include inorganic fibers such as glass fiber, rock wool, ceramic fiber alumina fiber, potassium titanate fiber or carbon fiber; or organic fibers such as an aromatic polyamide.

For improving properties of the resin in a final product such as a coating film, adhesive layer or molded article, a synthetic resin can be added and blended therewith. This type of synthetic resin is, for example, a thermosetting resin such as a phenol resin, an epoxy resin, a melamine resin or a silicone resin; polyamide; polycarbonate; polysulfone; polyether sulfone; polyether ether ketone; modified polyphenylene oxide; polyphenylene sulfide; polyether imide; or a fluorine-containing resin.

The thermosetting resin according to the present invention can be molded by known molding methods such as compression molding, transfer molding, extrusion molding or injection molding, and then effectively put to practical use.

The present invention will be described in further detail by reference to the following examples. However, these examples should not be construed as in any way limiting the invention.

#### Example 1A

In a reactor equipped with a stirrer, a reflux condenser, a thermometer, a Dean/Stark water separator and a nitrogen inlet pipe, there were placed 1410 g of N,N-dimethylformamide from which water had been removed, 207.6 g (0.6 mole) of 1,4-bis(4-hydroxy- $\alpha,\alpha$ -dimethylbenzyl)benzene, 193.7 g (1.23 mole) of 4-chloronitrobenzene and 248.8 g (1.8 moles) of potassium carbonate, and 50 g of toluene was then added thereto.

The resulting mixture was heated, and stirring was continued for 5 hours, while a temperature of from 140 to 150°C was maintained, in order to carry out the reaction. Water produced by the reaction was removed from the system by azeotropic distillation with toluene.

After completion of the reaction, the resulting inorganic salts were removed by hot filtration. The filtrate was then heated again. While a temperature of from 90 to 95°C was maintained, 210 g of water was added thereto over 2 hours in order to precipitate 1,4-bis[4-(4-nitrophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene. This solution containing the precipitate was slowly cooled, and the resulting light yellow crystals were collected by filtration and then washed with a mixed solvent of N,N-dimethylformamide and methanol. Afterward, the resulting sludge was washed with 1 liter of methanol, filtered and then dried, thereby obtaining 335 g of light yellow powder.

Using high-speed liquid chromatography, the purity of the product was determined as being 99.3%.

The analytical results were as follows:

Melting point: 186.5 to 188.5°C

#### Values of elemental analysis:

	C	H	N
Calcd. (%)	73.47	5.44	4.76
Found (%)	73.28	5.56	4.96

IR (KBr tablet method)

1330, 1500  $\text{cm}^{-1}$  (a nitro group)

1240  $\text{cm}^{-1}$  (an ether bond)

In a hermetic reduction reactor equipped with a stirrer and a thermometer there were placed 294 g (0.5 mole) of 1,4-bis[4-(4-nitrophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene obtained from the preceding step, 1175 g of dimethylformamide and 17.5 g of a 5% Pd/C catalyst. Hydrogen gas was then introduced therein, while the solution was stirred vigorously.

Reaction was continued at a reaction temperature of from 30 to 40°C for 4 hours. At this point, 67.2 liters of hydrogen had been absorbed by the solution. Since further absorption was not observed, the reaction was brought to an end.

After completion of the reaction, filtration was effected at room temperature to remove the Pd/C catalyst from the solution.

The resulting reaction solution was heated to 80 to 90°C. While this temperature was maintained, 500 g of water was added dropwise thereto over 2.5 hours in order to precipitate 1,4-bis[4-(4-aminophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene. This reaction solution was slowly cooled, and the resulting white crystals were collected by filtration and then washed with a mixed solution of N,N-dimethylformamide and methanol and afterward with methanol, and dried, thereby obtaining 242.8 g of 1,4 bis[4-(4-aminophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene.

Using high-speed liquid chromatography, the purity of the product was determined as being 99.2% (total yield 88.3%). The analytical results were as follows:

Melting point: 189 to 190.5°C

#### Values of elemental analysis:

	C	H	N
Calcd. (%)	81.82	6.82	5.30
Found (%)	81.90	6.75	5.21



IR (KBr tablet method)

1620, 3320-3430  $\text{cm}^{-1}$  (an amino group)

1230  $\text{cm}^{-1}$  (an ether bond)

# 5 EXAMPLE 1B

In a reactor equipped with a stirrer, a reflux condenser, a thermometer, a Dean/Stark water separator and a nitrogen inlet pipe, there were placed 600 g of N,N-dimethylformamide from which water had been removed, 207.6 g (0.6 mole) of 1,3-bis(4-hydroxy- $\alpha,\alpha$ -dimethylbenzyl)benzene, 193.7 g (1.23 mole) of 4-chloronitrobenzene and 248.8 g (1.8 moles) of potassium carbonate, and 50 g of toluene was then added thereto.

The resulting mixture was heated, and stirring was then continued for 5 hours, while a temperature of from 140 to 150°C was maintained, in order to carry out the reaction. Water produced by the reaction was removed from the system by azeotropic distillation with toluene.

After completion of the reaction, the resulting inorganic salts were removed by hot filtration. The filtrate was then heated again. While a temperature of from 90 to 95°C was maintained, 260 g of water was added thereto over 2 hours in order to precipitate 1,3-bis[4-(4-nitrophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene. This solution containing the precipitate was slowly cooled, and the resulting light yellow crystals were collected by filtration and then washed with a mixed solvent of N,N-dimethylformamide and methanol. Afterward, the resulting sludge was washed with 1 liter of methanol, filtered and then dried, thereby obtaining 328 g of light yellow powder.

Using high-speed liquid chromatography, the purity of the product was determined as being 99.1%. The analytical results were as follows:

Melting point: 154.5 to 156.0°C

## 25 Values of elemental analysis:

		C	H	N
	Calcd. (%)	73.47	5.44	4.76
30	Found (%)	73.34	5.20	4.79

IR (KBr tablet method)

1330, 1490  $\text{cm}^{-1}$  (a nitro group)

1230  $\text{cm}^{-1}$  (an ether bond)

35 In a reactor equipped with a stirrer, a thermometer, a reflux condenser and a dropping funnel, there were placed 294 g (0.5 mole) of 1,3-bis[4-(4-nitrophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene obtained from the preceding step. This material was then mixed with 1500 g of methyl cellosolve, 29.4 g of active carbon and 2.9 g of ferric chloride-hexahydrate. Afterward, stirring was continued for 3 hours, while a temperature of from 100 to 105°C was maintained.

40 While the temperature was maintained, 150.2 g of 80% hydrazine monohydrate was added dropwise over 3 hours. Furthermore, the solution was aged at this temperature for 1 hour. Solids were then removed by hot filtration. Afterward, the resulting filtrate was concentrated and then recrystallized with isopropyl alcohol. White crystals of 1,3-bis[4-(4-aminophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene were obtained which was the desired compound.

45 The white crystals were then filtered, washed with isopropyl alcohol, and dried to obtain 228.5 g of 1,3-bis[4-(4-aminophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene.

Using high-speed liquid chromatography, the purity of the product was determined as being 99.0% (total yield 81.4%).

The analytical results were as follows:

50 Melting point: 103 to 105.5°C

## Values of elemental analysis:

		C	H	N
55	Calcd. (%)	81.82	6.82	5.30
	Found (%)	81.86	6.45	5.22

IR (KBr tablet method)

1620, 3340-3440  $\text{cm}^{-1}$  (an amino group)

1240  $\text{cm}^{-1}$  (an ether bond)

#### 5 Example 1C

In a reactor equipped with a stirrer, a reflux condenser, a thermometer, a Dean/Stark water separator and a nitrogen inlet pipe, there were placed 400 g of N, N-dimethylformamide from which water had been removed, 161.0 g (0.4 mole) of 1,4-bis(4-hydroxy-3,5-dimethyl- $\alpha,\alpha$ -dimethylbenzyl)benzene, 138.7 g (0.88 mole) of 4-chloronitrobenzene and 66.3 g (0.48 mole) of potassium carbonate, and 50 g of toluene was then added thereto.

The resulting mixture was heated, and stirring was then continued for 5 hours, while a temperature of from 140 to 150°C was maintained, in order to carry out the reaction. Water produced by the reaction was removed from the system by azeotropic distillation with toluene.

After completion of the reaction, the resulting inorganic salts were removed by hot filtration. The filtrate was cooled in to precipitate 1,4-bis[4-(4-nitrophenoxy)-3,5-dimethyl- $\alpha,\alpha$ -dimethylbenzyl]benzene.

The precipitated light yellow crystals were collected by filtration, washed with 500 ml of methanol, and then dried. 245 g of light yellow powder was obtained.

Using high-speed liquid chromatography, the purity of the product was determined as 99.5%. The analytical results were as follows:

Melting point: 224 to 226°C

#### Values of elemental analysis:

		C	H	N
25	Calcd. (%)	74.51	6.25	4.34
	Found (%)	73.94	5.92	4.54

IR (KBr tablet method)

1345, 1515  $\text{cm}^{-1}$  (a nitro group)

1250  $\text{cm}^{-1}$  (an ether bond)

In a hermetic reduction reactor equipped with a stirrer and a thermometer there were placed 129.0 g (0.2 mole) of 1,4-bis[4-(4-nitrophenoxy)-3,5-dimethyl- $\alpha,\alpha$ -dimethylbenzyl]benzene obtained from the preceding step, 322 g of N,N-dimethylformamide and 2.6 g of a 5% Pd/C catalyst. Hydrogen gas was then introduced thereinto, while the solution was stirred vigorously.

Reaction was continued at a reaction temperature of from 30 to 40°C for 4 hours. At this point, 26.9 liters of hydrogen had been absorbed by the solution. Since further absorption was not observed the reaction was brought to an end.

After completion of the reaction, filtration was effected at room temperature to remove the Pd/C catalyst from the solution.

The resulting reaction solution was heated to 80 to 90°C. While this temperature was maintained, 80 g of water was added dropwise thereto over 2 hours in order to precipitate 1,4-bis[4-(4-aminophenoxy)-3,5-dimethyl- $\alpha,\alpha$ -dimethylbenzyl]benzene. The reaction solution was slowly cooled, and the resulting light yellow crystals were collected by filtration, washed with 200 ml of methanol, and then dried, thereby obtaining 104.7 g of 1,4-bis[4-(4-aminophenoxy)-3,5-dimethyl- $\alpha,\alpha$ -dimethylbenzyl]benzene.

Using high-speed liquid chromatography, the purity of the product was determined as being 99.2 % (total yield 85.0%). The analytical results were as follows:

Melting point: 228 to 230°C

#### 50 Values of elemental analysis:

		C	H	N
	Calcd. (%)	82.15	7.58	4.79
55	Found (%)	82.09	7.13	4.84

IR (KBr tablet method)

1630, 3320-3430  $\text{cm}^{-1}$  (an amino group)

1240  $\text{cm}^{-1}$  (an ether bond)

#### Example 1D

In a reactor equipped with a stirrer, a reflux condenser, a thermometer, a Dean/stark water separator and a nitrogen inlet pipe, there were placed 400 g of N,N-dimethylformamide from which water had been removed, 138.4 g (0.4 mole) of 1,4-bis(4-hydroxy- $\alpha,\alpha$ -dimethylbenzyl)benzene, 147.8 g (0.88 mole) of m-dinitrobenzene and 66.3 g (0.48 mole) of potassium carbonate, and 50 g of toluene was then added thereto.

The resulting mixture was heated, and stirring was then continued for 5 hours, while a temperature of from 140 to 150°C was maintained, in order to carry out the reaction. Water produced by the reaction was removed from the system by azeotropic distillation with toluene.

After completion of the reaction, the resulting inorganic salts were removed by hot filtration. Then, the filtrate was heated again. While a temperature of from 90 to 95°C was maintained, 112 g of water was added thereto over 2 hours in order to precipitate 1,4-bis[4-(3-nitrophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene. This solution was slowly cooled, and the resulting faintly brown crystals were collected by filtration, washed with 500 ml of methanol, and then dried. 220 g of faintly brown powder was obtained.

Using high-speed liquid chromatography, the purity of the product was determined as being 99.3%. The analytical results were as follows:

Melting point: 123 to 125°C

#### Values of elemental analysis:

	C	H	N
Calcd. (%)	73.45	5.48	4.76
Found (%)	73.53	5.39	4.74

IR (KBr tablet method)

1355, 1525  $\text{cm}^{-1}$  (a nitro group)

1240  $\text{cm}^{-1}$  (an ether bond)

In a hermetic reduction reactor equipped with a stirrer and a thermometer, there were placed 118 g (0.2 mole) of 1,4-bis[4-(3-nitrophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene obtained from the preceding step, 295 g of N,N-dimethyl-formamide and 2.4 g of a 5% Pd/C catalyst. A hydrogen gas was then introduced thereinto, while the solution was stirred vigorously.

Reaction was continued at a reaction temperature of from 30 to 40°C for 4 hours. At this point, 26.9 liters of hydrogen had been absorbed by the solution. Since further absorption was not observed, the reaction was brought to an end.

After completion of the reaction, filtration was effected at room temperature to remove the Pd/C catalyst from the solution.

The resulting reaction solution was heated to 80 to 90°C. While this temperature was maintained, 126 g of water was added dropwise thereto over 2 hours in order to precipitate 1,4-bis[4-(3-aminophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene. The reaction solution was slowly cooled, and the resulting light yellow crystals were collected by filtration, washed with 200 ml of methanol, and then dried, thereby obtaining 90.3 g of 1,4-bis[4-(3-aminophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene.

Using high-speed liquid chromatography, the purity of the product was determined as being 99.2% (total yield 80.0%).

The analytical results were as follows:

Melting point: 138 to 140°C

#### Values of elemental analysis:

	C	H	N
Calcd. (%)	81.79	6.86	5.30
Found (%)	81.43	6.79	5.27

IR (KBr tablet method)

1630, 3320-3430  $\text{cm}^{-1}$  (an amino group)

1240  $\text{cm}^{-1}$  (an ether bond)

### Example 1E

In a reactor equipped with a stirrer, a reflux condenser, a thermometer, a Dean/Stark water separator and a nitrogen inlet pipe, there were placed 400 g of N,N-dimethylformamide from which water had been removed, 138.6 g (0.4 mole) of 1,3-bis(4-hydroxy- $\alpha,\alpha$ -dimethylbenzyl)benzene, 147.8 g (0.88 mole) of m-dinitrobenzene and 66.3 g (0.48 mole) of potassium carbonate, and 50 g of toluene was then added thereto.

The resulting mixture was heated, and stirring was then continued for 5 hours, while a temperature of from 140 to 150°C was maintained, in order to carry out the reaction. Water produced by the reaction was removed from the system by azeotropic distillation with toluene.

After completion of the reaction, the resulting inorganic salts were removed by filtration. Then, the filtrate was heated again. While a temperature of from 90 to 95°C was maintained, 112 g of water was added thereto over 2 hours in order to precipitate 1,3-bis[4-(3-nitrophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene. This solution was slowly cooled, and the resulting faintly brown crystals were collected by filtration, washed with 200 ml of methanol, and then dried. 215 g of faintly brown powder was obtained.

Using high-speed liquid chromatography, the purity of the product was determined as being 99.1%. The analytical results were as follows;

Melting point: 84 to 86°C

#### Values of elemental analysis:

		C	H	N
Calcd. (%)		73.45	5.48	4.76
Found (%)		73.70	5.40	4.46

IR (KBr tablet method)

1330, 1510  $\text{cm}^{-1}$  (a nitro group)

1245  $\text{cm}^{-1}$  (an ether bond)

In a hermetic reduction reactor equipped with a stirrer and a thermometer, there were placed 118 g (0.2 mole) of 1,3-bis[4-(3-nitrophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene obtained from the preceding step, 295 g of N,N-dimethylformamide and 2.4 g of a 5% Pd/C catalyst. Hydrogen gas was then introduced thereto, while the solution was stirred vigorously.

Reaction was continued at a reaction temperature of from 30 to 40°C for 5 hours. At this point, 26.9 liters of hydrogen had been absorbed by the solution. Since further absorption was not observed any more, the reaction was brought to an end.

After completion of the reaction, filtration was effected at room temperature to remove the Pd/C catalyst from the solution.

The resulting reaction solution was heated to 80 to 90°C. While this temperature was maintained, 126 g of water was added dropwise thereto over 2 hours in order to precipitate 1,3-bis[4-(3-aminophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene. The reaction solution was slowly cooled, and the resulting light yellow crystals were collected by filtration, recrystallized from methanol, and then dried, thereby obtaining 86.6 g of 1,3-bis[4-(3-aminophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene.

Using high-speed liquid chromatography, the purity of the product was determined as being 99.0% (total yield 75.0%).

The analytical results were as follows:

Melting point: 94 to 96°C

#### Values of elemental analysis:

		C	H	N
Calcd. (%)		81.82	6.82	5.30
Found (%)		81.35	6.96	5.44

IR (KBr tablet method)

1625, 3340-3440  $\text{cm}^{-1}$  (an amino group)1235  $\text{cm}^{-1}$  (an ether bond)5 Example 2A

In a reactor equipped with a stirrer, a thermometer, a reflux condenser having a water separator and a dropping funnel there were placed 26.5 g (0.27 mole) of maleic anhydride, 1.6 g of p-toluenesulfonic acid and 150 ml of toluene, and the temperature of the solution was increased to maintain the reflux condition of toluene. A solution prepared by dissolving 52.8 g (0.1 mole) of 1,3-bis[4-(4-aminophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene formed in Example 1B in 100 ml of toluene was added dropwise to the above-mentioned solution through a dropping funnel over 7 hours. After completion of the addition, the solution was aged for 2 hours. Water produced by the reaction was collected by a water separator provided on the reflux condenser from the beginning of the addition to the end of the aging. After completion of the reaction, the solution was cooled to 70°C, and 100 ml of warm water was added thereto. Stirring followed at this temperature for 30 minutes. After standing, the lower layer (the water layer) was discarded, and 100 ml of warm water was again added. Washing and liquid separation followed in like manner.

The resulting upper layer (toluene solution) was finally concentrated in vacuum at a temperature of not more than 130°C to obtain 68 g of reddish brown transparent resinous crude 1,3-bis[4-(4-maleimidophenoxy)- $\alpha,\alpha$ -dimethylbenzyl] benzene which was the desired compound. Its yield was quantitative, and its purity according to HLC was 91%.

Afterward, the crude crystals were purified by recrystallization to obtain pure light yellow powder. The analytical results were as follows:

Melting point: 172 to 175°C

25

Values of elemental analysis ( $\text{C}_{44}\text{H}_{36}\text{N}_2\text{O}_6$ ):

		C	H	N
30	Calcd. (%)	76.70	5.27	4.07
	Found (%)	76.20	5.43	4.22

A mass spectrum is shown in Fig. 1, and an IR spectrum is shown in Fig. 2.

35 Example 2B

In the same reactor used in Example 2A, there were placed 26.5 g (0.27 mole) of maleic anhydride, 0.7 g of phosphoric acid and 150 ml of mixed xylenes. A solution was then added dropwise thereto under reflux of xylene, the solution having been prepared by dissolving 52.8 g (0.1 mole) of 1,4-bis[4-(4-aminophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene obtained in Example 1A in a mixture of 100 ml of mixed xylene and 50 ml of N,N-dimethylformamide, with heating. An addition time of 5 hours and an aging time of 2 hours were used in order to carry out the reaction. The reaction solution was concentrated to about 1/2 the original amount and then cooled. When a temperature of 80°C had been reached, 50 g of water was added. Afterward, the solution was slowly cooled with stirring, so that crystals were precipitated. The crystals were then cooled to room temperature, filtered, and washed with cold methanol. 66.5 g of desired crude 1,4-bis[4-(4-maleimidophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene was obtained. Its yield was 96.7%, and its purity according to HLC was 97.5%.

The product was then purified by recrystallization from toluene in order to obtain pure light yellow powder. The analytical results were as follows:

Melting point: 214 to 216°C

50

Values of elemental analysis ( $\text{C}_{44}\text{H}_{36}\text{N}_2\text{O}_6$ ):

		C	H	N
55	Calcd. (%)	76.70	5.27	4.07
	Found (%)	76.02	5.51	4.18

The results of a mass spectrum were the same as for the compound of Example 2A (M/Z; M+ 688, 673,

408).

Experiment 1

With regard to the bismaleimides obtained in Examples 2A and 2B, their solubility in some solvents were investigated. The results are set forth in Table 1.

Table 1  
Solubility of Bismaleimide in Solvent\*  
(g/100 ml)

Solvent	Compound of Example 2A	Compound of Example 2B	N,N' (Methylene- di-P-phenylene)- bismaleimide
Toluene	>50	8	<1
Tetra- hydro- furan	>50	12	6
Methylene Chloride	<50	15	8

\* Measurement was made at 25°C.

Example 2C

In a reactor equipped with a stirrer, a thermometer, a reflux condenser having a water separator and a dropping funnel there were placed 26.5 g (0.27 mole) of maleic anhydride, 1.6 g of p-toluenesulfonic acid and 150 ml of toluene. The temperature of the solution was increased to cause the toluene to boil under reflux. A solution prepared by dissolving 58.8 g (0.1 mole) of 1,4-bis[4-(4-aminophenoxy)-3,5-dimethyl- $\alpha,\alpha$ -dimethylbenzyl]benzene formed in Example 1C in 100 ml of toluene was added dropwise to the above-mentioned solution through a dropping funnel over 7 hours. After completion of the addition, the solution was aged for 2 hours. Water produced by the reaction was collected by a water separator provided on the reflux condenser from the beginning of the addition to the end of the aging. After completion of the reaction, the solution was cooled to 70°C, and 100 ml of warm water was added thereto. Stirring followed at this temperature for 30 minutes. After standing, the lower layer (the water layer) of two separate layers was discarded, and 100 ml of warm water was again added. Washing and liquid separation followed in like manner.

The resulting upper layer (toluene solution) was cooled in order to precipitate light yellow crude 1,4-bis[4-(4-maleimidophenoxy)-3,5-dimethyl- $\alpha,\alpha$ -dimethylbenzyl]benzene, which was collected by filtration and then dried. 65.2 g of product was obtained.

Its yield was 87.5%, and its purity according to HLC was 96.4%.

Afterward, the crude crystals were purified by recrystallization to obtain a pure light yellow powder. The analytical results were as follows:

Melting point: 232.5 to 235°C

Values of elemental analysis ( $C_{48}H_{44}N_2O_6$ ):

		C	H	N
	Calcd. (%)	77.38	5.95	3.76
5	Found (%)	77.05	6.08	3.66

An IR spectrum of the product is shown in Fig. 3.

10 Example 2D

In a reactor equipped with a stirrer, a thermometer, a reflux condenser having a water separator and a dropping funnel there were placed 26.5 g (0.27 mole) of maleic anhydride, 1.6 g of p-toluenesulfonic acid and 150 ml of toluene. The temperature of the solution was increased to maintain the toluene under reflux. A solution prepared by dissolving 52.8 g (0.1 mole) of 1,3 bis[4-(3-aminophenoxy)- $\alpha,\alpha$ -dimethyl-benzyl]benzene formed in Example 1E in 100 ml of toluene was added dropwise to the above-mentioned solution through a dropping funnel over 7 hours. After completion of the addition, the solution was aged for 2 hours. Water produced by the reaction was collected by a water separator provided on the reflux condenser from the beginning of the addition to the end of the aging. After completion of the reaction, the solution was cooled to 70°C, and 100 ml of warm water was added thereto. Stirring followed at this temperature for 30 minutes. After standing, the lower layer (the water layer) of two separate layers was discarded, and 100 ml of warm water was again added. Washing and liquid separation followed in like manner.

The resulting upper layer (toluene solution) was finally concentrated in vacuo at a temperature of not more than 130°C to obtain 68.2 g of reddish brown transparent resinous crude 1,3-bis[4-(3-maleimidophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene. This was the desired compound.

Its yield was quantitative, and its purity according to MLC was 96.3%.

Afterward, the crude crystals were purified by column chromatography to obtain a pure light yellow resinous product.

30 Elemental analysis ( $C_{44}H_{36}N_2O_6$ ):

		C	H	N
	Calcd. (%)	76.70	5.27	4.07
35	Found (%)	76.41	5.44	4.01

Mass spectrum

M/Z;  $M^+$  688, 673, 408

An IR spectrum of the product is shown in Fig. 4.

40 Example 2E

In the same reactor used in Example 2D, there were placed 26.5 g (0.27 mole) of maleic anhydride, 0.7 g of phosphoric acid and 150 ml of mixed xylenes. A solution was then added dropwise thereto, the solution having been prepared by dissolving with heating 52.8 g (0.1 mole) of 1,4-bis[4-(3-aminophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene obtained in Example 1D in a mixture of 100 ml of mixed xylenes and 50 ml of N,N-dimethylformamide. An addition time of 5 hours and an aging time of 2 hours were used in order to carry out the reaction. The reaction solution was concentrated to about 1/2 the original amount and then cooled. When a temperature of 80°C had been reached, 50 g of water was added thereto. Afterward, the solution was slowly cooled, so that crystals precipitated. The precipitated crystals were then cooled to room temperatures filtered; and washed with cold methanol. 67.1 g of desired crude 1,4-bis[4-(3-maleimidophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene was obtained.

Its yield was 94.8%, and its purity according to HLC was 92.0%.

The product was then purified by recrystallization from toluene in order to obtain pure light yellow powder.

55 The analytical results were as follows:

Melting point: 192 to 193°C

Elemental analysis ( $C_{44}H_{36}N_2O_6$ ):

	C	H	N
Calcd. (%)	76.70	5.27	4.07
Found (%)	76.66	5.33	4.18

The results of a mass spectrum were the same as from the compound from Example 2D (M/Z; M<sup>+</sup> 688, 673, 408).

#### Examples 3A to 3C and Comparative Examples 3A and 3B

In a reactor equipped with a stirrer, a reflux condenser and a nitrogen inlet pipe, there were placed 1,3-bis[4-(4-maleimidophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene obtained in Example 2A and 1,3-bis[4-(4-aminophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene obtained in Example 1B in the molar ratio set forth in Table 2. They were then heated and melted at 180°C for 20 minutes. Then, they were defoamed at 150°C under reduced pressure (from 10 to 15 mmHg) (1300 - 1950 Nm<sup>-2</sup>) for 30 minutes. Afterward, the mixture was cooled to room temperature, thereby to obtain a brown transparent glassy solidified resin forming composition. A mold which had been heated to 180°C was filled with the resin forming composition, while the resin composition was heated and melted. The composition was maintained at 200°C under 50 kg/cm<sup>2</sup> for 30 minutes. After this compression molding, the molded material was taken out of the mold and postcured in an oven at 250°C for 4 hours. Cured test pieces (length 127 mm, width 12.7 mm and thickness 6.4 mm) were obtained.

For these test pieces, bending strength, flexural modulus, the starting temperature of thermal decomposition and hygroscopicity were measured, and the results are set forth in Table 2.

#### COMPARATIVE EXAMPLE 3C

N,N',4,4'-Diphenylmethanebismaleimide and 4,4'-diaminodiphenylmethane were used to prepare a composition (molar ratio 2/1) shown in Table 2. Afterward, the same operation explained with respect to Examples 3A to 3C was followed, and the results set forth in Table 2 were obtained.

#### COMPARATIVE EXAMPLE 3D

Kelimide 601 (made by Nippon Polyimide Co., Ltd.) was used as a resin forming composition. afterward, the same operation explained with respect to Examples 3A to 3C was followed, and the results set forth in Table 2 were obtained.

#### EXAMPLES 3D TO 3G

In a stainless steel reactor equipped with a stirrer, a reflux condenser and a nitrogen inlet pipe, there were placed each of the bismaleimide compounds prepared in Examples 2A, 2D, 2C and 2E and each of the diamine compounds prepared in Examples 1A, 1C and 1B in the molar ratios shown in Table 3. N-methyl-2-pyrrolidone was poured into the solution in such an amount that the concentration of a resin was 55% by weight, followed by heating at 150°C for 50 minutes. The resulting varnish solution was discharged into water to form a precipitate. The precipitate was then collected by filtration, washed with water, and then dried with hot air at 80°C for 15 hours. The product was further dried at 110°C for 20 minutes, and additionally at 130°C for 30 minutes. It was then ground in a mortar and then passed through a sieve of 60 mesh, to obtain a polyaminobismaleimide type thermosetting resin forming composition. Afterward, the same operation described with respect to Examples 3A to 3C was followed, and the results set forth in Table 3 were obtained.

In these examples, physical properties of thermosetting resins and the like were measured in the following manner.

Bending strength and flexural modulus were measured in accordance with ASTM-D-790.

Starting temperature of thermal decomposition was determined as the temperature at which weight begins to decrease and was measured at a temperature rise rate of 10°C/minute in air in accordance with the TGA method.

Izod impact strength was measured in accordance with ASTM-D 256.

Hygroscopicity was measured in accordance with ASTM-D570-63.



Table 2

Resin Forming Composition	Bismaleimide Compound (IV)	Diamine Compound (I)
Example 3A	1,3-bis(4-(4-maleimidophenoxy)- a,a-dimethylbenzyl)benzene	1,3-bis[4(4-aminophenoxy)- a,a-dimethylbenzyl]benzene
Example 3B	ditto	ditto
Example 3C	ditto	ditto
Comp. Ex. 3A	ditto	ditto
Comp. Ex. 3B	ditto	ditto
Comp. Ex. 3C	N,N',4,4'-diphenyl- methanebismaleimide	4,4'-diaminodiphenylmethane
Comp. Ex. 3D	Kellamide 601	

Table 2 (Continued)

	Molar ratio I/IV	Izod impact strength (no notch) (kg.cm/cm)	Bending strength (kg/mm <sup>2</sup> )	Flexural modulus (kg/mm <sup>2</sup> )	Temp. at 0.5% weight decrease (°C)	Hygro- scopicity (%)
Example 3A	0.3	37	15.0	318	360	0.26
Example 3B	0.5	41	16.1	316	357	0.29
Example 3C	1.0	38	15.4	317	362	0.30
Comp. Ex. 3A	1.5	37	15.8	305	315	0.27
Comp. Ex. 3B	0	2	5.2	460	412	0.90
Comp. Ex. 3C	0.5	11	9.2	372	359	0.89
Comp. Ex. 3D		9	8.1	342	366	0.90

Table 3

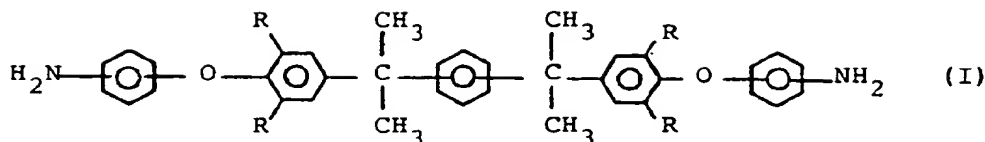
R e s i n F o r m i n g C o m p o s i t i o n		
	Bismaleimide Compound (IV)	Diamine Compound (I)
Example 3D	1,3-bis[4-(4-maleimidophenoxy)- α,α-dimethylbenzyl]benzene	1,4-bis[4-(4-aminophenoxy)- α,α-dimethylbenzyl]benzene
Example 3E	1,3-bis[4-(3-maleimidophenoxy)- α,α-dimethylbenzyl]benzene	1,4-bis[4-(4-aminophenoxy)- 3,5-dimethyl-α,α-dimethyl- benzyl]benzene
Example 3F	1,4-bis[4-(4-maleimidophenoxy)- 3,5-dimethyl-α,α-dimethyl- benzyl]benzene	1,3-bis[4-(4-aminophenoxy)- α,α-dimethylbenzyl]benzene
Example 3G	1,4-bis[4-(3-maleimidophenoxy)- α,α-dimethylbenzyl]benzene	1,3-bis[4-(4-aminophenoxy)- α,α-dimethylbenzyl]benzene

Table 3 (Continued)

	Molar ratio I/IV	Izod impact strength (no notch) (kg.cm/cm)	Bending strength (kg/mm <sup>2</sup> )	flexural modulus (kg/mm <sup>2</sup> )	Temp. at 0.5% weight decrease (°C)	Hygro- scopicit (%)
Example 3D	0.5	37	15.5	325	362	0.27
Example 3E	0.5	35	15.7	320	359	0.24
Example 3F	0.5	35	15.0	340	360	0.29
Example 3G	0.5	36	14.8	332	356	0.26

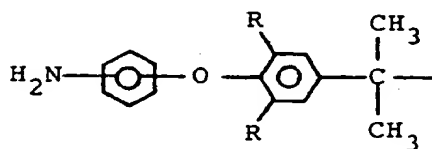
## Claim

1. An aromatic diamine compound having a structure represented by formula (I):



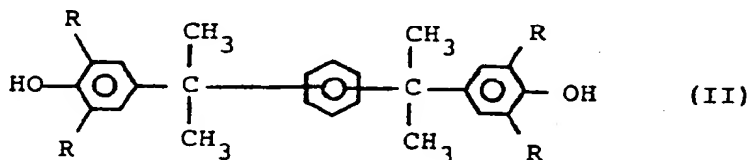
wherein the R groups may be the same or different and are H or CH<sub>3</sub>.

2. A compound according to Claim 1, wherein the amino groups of the aminophenoxy groups are both bonded at meta positions or both are bonded at para positions.
3. A compound according to Claim 1, wherein the two groups represented by the formula:

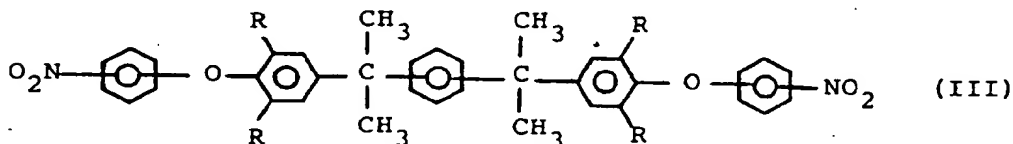


are both bonded at meta positions, or both are bonded at para positions.

4. A compound according to Claim 1 selected from 1,4- and 1,3-bis[4-(4-amino-phenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene, 1,4- and 1,3-bis[4-(3-aminophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene, 1,3-bis[4-(3-aminophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene, 1,4- and 1,3-bis[4-(4-aminophenoxy)-3,5-dimethyl- $\alpha,\alpha$ -dimethylbenzyl]benzene, and 1,4- and 1,3-bis[4-(3-aminophendxy)-3,5-dimethyl- $\alpha,\alpha$ -dimethylbenzyl]benzene.
5. A method for preparing a compound according to any preceding claim, said method comprising (i) reacting a compound having a structure represented by formula (II):



wherein the R groups may be the same or different and are H or CH<sub>3</sub>, with a nitrobenzene compound selected from nitrochlorobenzenes and dinitrobenzenes, in a non-protonic polar solvent, in the presence of a base, to form a compound having a structure represented by formula (III):

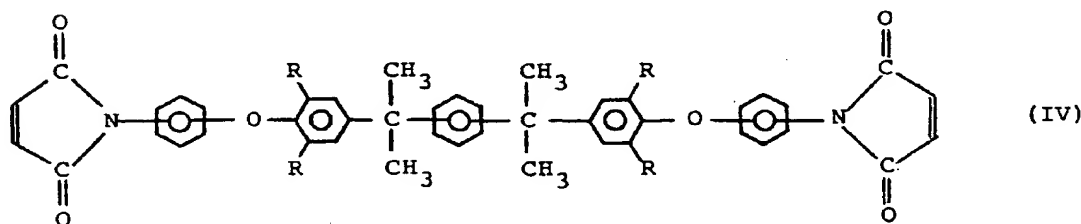


wherein the R groups may be the same or different and are H or CH<sub>3</sub>, and then (ii) reducing the compound having a structure represented by formula (III).

6. A method according to Claim 5 wherein the nitrobenzene compound is m-dinitrobenzene and the compound having a structure represented by formula (III) is reduced to produce a compound having a structure represented by formula (I) wherein the amino groups of the aminophenoxy groups are bonded at meta positions; or wherein

the nitrobenzene compound is 4-chloronitrobenzene, and the compound having a structure represented by formula (III) is reduced to produce a compound having a structure represented by formula (I) wherein the amino groups of the aminophenoxy groups are bonded at para positions.

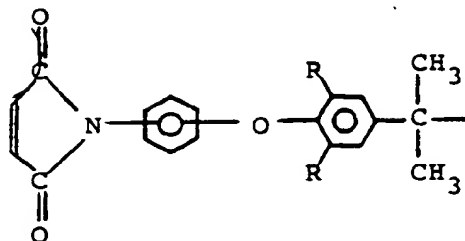
7. A method according to Claim 5 wherein : ((i) the compound having a structure of formula (II) is selected from 1,4- and 1,3-bis-(4-hydroxy- $\alpha,\alpha$ -dimethylbenzyl)benzene, the nitrobenzene compound is 4-chloronitrobenzene, and wherein 1,4- or 1,3-bis[4-(4-amino-phenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene is obtained ; or  
 (ii) the compound having a structure represented by formula (II) is selected from 1,4- and 1,3-bis(4-hydroxy- $\alpha,\alpha$ - dimethylbenzyl)benzene, the nitrobenzene compound is m-dinitrobenzene, and wherein 1,4- or 1,3-bis[4-(3-aminophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene is obtained; or  
 (iii) the compound having a structure represented by formula (II) is selected from 1,4- and 1,3-bis-(4-hydroxy-3,5-dimethyl- $\alpha,\alpha$ -dimethylbenzyl)benzene, the nitrobenzene compound is 4-chloronitrobenzene, and wherein 1,4- or 1,3-bis[4-(4-aminophenoxy)-3,5-dimethyl- $\alpha,\alpha$ -dimethylbenzyl]benzene is obtained; or  
 (iv) the compound having a structure represented by formula (II) is selected from 1,4- and 1,3-bis(4-hydroxy-3,5-dimethyl- $\alpha,\alpha$ -dimethylbenzyl)benzene, the nitrobenzene compound is m-dinitrobenzene, and wherein 1,4- or 1,3-bis[4-(3-aminophenoxy)-3,5-dimethyl- $\alpha,\alpha$ -dimethylbenzyl]benzene is obtained.
8. The method according to any of claims 5-7 using two to three moles of nitrobenzene compound per mole of said compound having a structure represented by formula (II).
9. A method according to any of claims 5-8 wherein said base is at least one compound selected from carbonates, hydrogen carbonates, hydroxides and alkoxides of alkaline metals.
10. A method according to any of claims 5-9, wherein said base is used in an amount of 2 equivalents or more based on the amount of said compound having a structure represented by formula (II).
11. A method according to any of claims 5-10, wherein said solvent is at least one of dimethyl sulfoxide, N,N-dimethylformamide, N,N'-dimethylacetamide, N-methylpyrrolidone, 1,3-dimethyl-2-imidazolidinone, hexamethylphosphortriamide and sulfolane.
12. A method according to any of claims 5-11, wherein said solvent is used in an amount which is 1 to 15 times the amount of said compound having a structure represented by formula (II) based on weight.
13. A method according to any of claims 5-12, wherein the reaction between the compound having a structure of formula (II) and the nitrobenzene compound is carried out in the range of from 50 to 220°C.
14. A method according to any of claims 5-13, wherein the compound having a structure represented by formula (III) is reduced by catalytic reduction or hydrazine reduction.
15. A bismaleimide compound having a structure represented by formula (IV):



wherein the R groups may be the same or different and are of H or CH<sub>3</sub>.

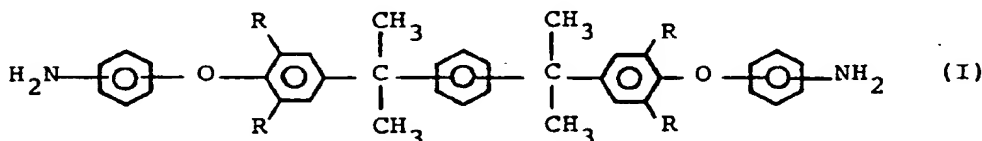
16. A compound according to claim 15 wherein the maleimido groups of the maleimidophenoxy groups are both bonded at meta positions or both bonded at para positions.

17. A compound according to claim 15 wherein the two groups  $r$  presented by the formula:



are both bonded at meta positions, or both are bonded at para positions.

18. The compound according to claim 15, selected from 1,4- and 1,3-bis[4-(4-maleimidophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene, 1,4- and 1,3-bis[4-(3-maleimidophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene, 1,4- and 1,3-bis[4-(4-maleimidophenoxy)-3,5-dimethyl- $\alpha,\alpha$ -dimethylbenzyl]benzene, and 1,4- and 1,3-bis[4-(3-maleimidophenoxy)-3,5-dimethyl- $\alpha,\alpha$ -dimethylbenzyl]benzene.
19. A method for preparing a compound according to any of claims 15-18 said method comprising reacting an aromatic diamine compound represented by formula (I):

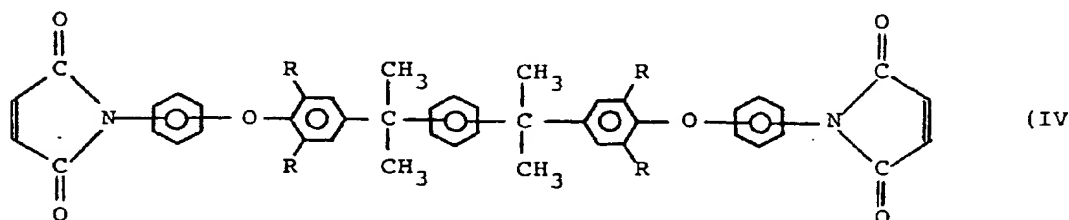


wherein the R groups may be the same or different and are H or  $\text{CH}_3$ , with maleic anhydride in the presence of an acid catalyst under heating/reflux of an organic solvent, while removing any water produced from the system.

20. A method according to Claim 19, wherein the aromatic diamine compound is selected from the group of 1,4- and 1,3-bis[4-(4-aminophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene, whereby 1,4- or 1,3-bis[4-(4-maleimidophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene is obtained;  
or from  
1,4- and 1,3-bis[4-(3-aminophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene, whereby 1,4- or 1,3-bis[4-(3-maleimidophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene is obtained;  
or from  
1,4- and 1,3-bis[4-(4-aminophenoxy)-3,5-dimethyl- $\alpha,\alpha$ -dimethylbenzyl]benzene, whereby 1,4- or 1,3-bis[4-(4-maleimidophenoxy)-3,5-dimethyl- $\alpha,\alpha$ -dimethylbenzyl]benzene is obtained;  
or from  
1,4- and 1,3-bis[4-(3-aminophenoxy)-3,5-dimethyl- $\alpha,\alpha$ -dimethylbenzyl]benzene, whereby 1,4- or 1,3-bis[4-(3-maleimidophenoxy)-3,5-dimethyl- $\alpha,\alpha$ -dimethylbenzyl]benzene is obtained.
21. A method according to claim 19 or 20 wherein said acid catalyst is at least one of mineral acids, heteropolyacids, organic sulfonic acids, halogenated carboxylic acids, solid acids, diamine salts thereof, and cation exchange type ion exchange resins,  
and is preferably  
at least one member selected from the group of sulfuric acid, hydrochloric acid, phosphoric acid, phosphotungstic acid, phosphomolybdic acid, p-toluenesulfonic acid, methanesulfonic acid, trichloroacetic acid, trifluoroacetic acid, silica-alumina solid acid, and diamine salts thereof.
22. A method according to claim 19, 20 or 21 wherein the acid catalyst is used in an amount of from 0.1 to 10% by weight based on the total weight of the aromatic diamine compound and maleic anhydride.
23. A method according to any of claims 19-22 wherein the organic solvent is at least one of aliphatic hydrocarbons, alicyclic hydrocarbons, aromatic hydrocarbons, halides of said aliphatic and aromatic hydrocar-

bons, and oxygen-containing, nitrogen-containing and sulfur-containing polar solvents.

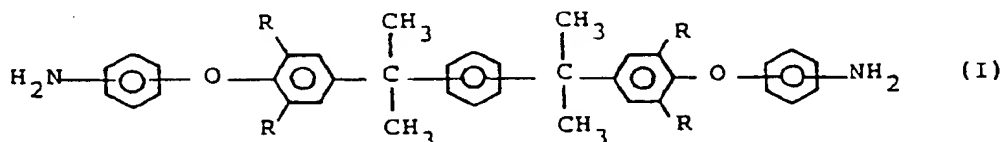
24. A method according to any of claims 19-23 wherein said organic solvent is used in an amount of 1 to 20 times as much as the total weight of said aromatic diamine and maleic anhydride.
25. The method according to any of claims 19-24 wherein the reaction is carried out at a temperature in the range of from 80 to 190°C.
26. A method according to any of claims 19-25 wherein the molar ratio of maleic anhydride to aromatic diamine compound is 2.05 to 3.
27. A resin forming composition comprising said aromatic diamine compound according to any of claims 1-4 and a bismaleimide compound having a structure represented by formula IV:



- wherein the R groups may be the same or different and are H or CH<sub>3</sub>, said aromatic diamine compound being present in a ratio of from 0.1 to 1.2 moles per mole of said bis-maleimide compound.
28. A resin forming composition comprising a bismaleimide compound according to any of claims 15-18 and an aromatic diamine compound selected from 1,4- and 1,3-bis[4-(4-aminophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene, 1,4- and 1,3-bis[4-(3-aminophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzene, 1,4- and 1,3-bis[4-(4-aminophenoxy)-3,5-dimethyl- $\alpha,\alpha$ -dimethylbenzyl]benzene, and 1,4- and 1,3-bis[4-(3-aminophenoxy)-3,5-dimethyl- $\alpha,\alpha$ -dimethylbenzyl]benzene; said aromatic diamine compound being present in a ratio of 0.1 to 1.2 moles per mole of said bismaleimide.
29. A resin of the resin forming composition according to claim 27 or 28.
30. A resin forming composition according to claim 27 or 28 further comprising at least one additive selected from a curing accelerator, a reinforcing material, a filler and a synthetic resin.
31. A method for preparing a resin composition according to claim 27 or 28 comprising (i) mixing said bismaleimide compound with said aromatic diamine compound, and then (ii) subjecting the resulting mixture to heat treatment at a temperature of from 70 to 220°C to carry out prepolymerization.
32. A method according to Claim 31 wherein the bismaleimide compound and aromatic diamine compound are mixed in the presence of an organic solvent.

#### Patentansprüche

1. Aromatische Diaminverbindung mit einer Struktur, die durch die Formel (I):

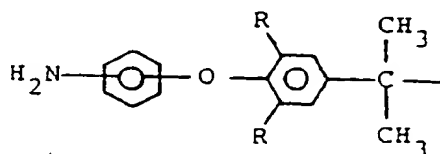




repräsentiert wird, worin die R-Gruppen gleich oder unterschiedlich sein können und H oder CH<sub>3</sub> sind.

2. Verbindung nach Anspruch 1, worin die Aminogruppen der Aminophenoxy-Gruppen beide in Meta-Stellungen oder beide in Para-Stellungen gebunden sind.

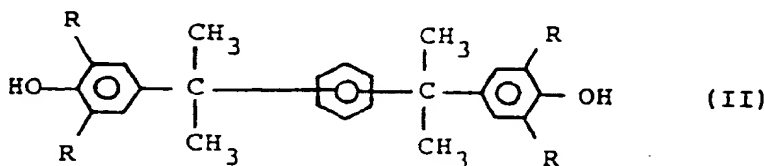
3. Verbindung nach Anspruch 1, worin die beiden durch die Formel



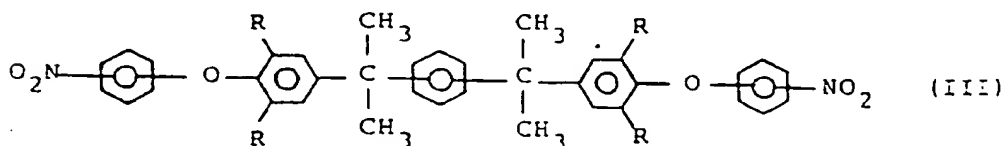
repräsentierten Gruppen beide in Meta-Stellungen oder beide in Para-Stellungen gebunden sind.

4. Verbindung nach Anspruch 1, ausgewählt aus 1,4- und 1,3-Bis[4-(4-amino-phenoxy)- $\alpha,\alpha$ -dimethylbenzyl]-benzol, 1,4- und 1,3-Bis[4-(3-aminophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]-benzol, 1,4- und 1,3-Bis[4-(4-aminophenoxy)-3,5-dimethyl- $\alpha,\alpha$ -dimethylbenzyl]-benzol und 1,4- und 1,3-Bis[4-(3-aminophenoxy)-3,5-dimethyl- $\alpha,\alpha$ -dimethylbenzyl]-benzol.

5. Verfahren zur Herstellung einer Verbindung nach einem der vorhergehenden Ansprüche, wobei das Verfahren umfaßt (i) die Reaktion einer Verbindung mit einer Struktur, die durch die Formel (II) repräsentiert wird



worin die R-Gruppen gleich oder unterschiedlich sein können und H oder CH<sub>3</sub> sind, mit einer Nitrobenzolverbindung, ausgewählt aus Nitrochlorbenzolen und Dinitrobenzolen in einem nicht-protonischen polaren Lösungsmittel in Anwesenheit einer Base zur Bildung einer Verbindung mit einer Struktur, die durch die Formel (III) repräsentiert wird



worin die R-Gruppen gleich oder unterschiedlich sein können und H oder CH<sub>3</sub> darstellen, und dann (ii) die Verbindung mit einer Struktur, die durch die Formel (III) dargestellt wird, der Reduktion unterworfen wird.

6. Verfahren nach Anspruch 5, worin die Nitrobenzolverbindung m-Dinitrobenzol ist und die Verbindung mit einer Struktur, die durch die Formel (III) repräsentiert wird, der Reduktion unterworfen wird, um eine Verbindung mit einer Struktur zu erzeugen, die durch die Formel (I) repräsentiert wird, worin die Aminogruppen der Aminophenoxygruppen in Meta-Stellung gebunden sind; oder worin die Nitrobenzolverbindung 4-Chlornitrobenzol ist und die Verbindung eine Struktur aufweist, die durch die Formel (III) repräsentiert wird und zur Erzeugung einer Verbindung mit einer Struktur, die durch die Formel (I) wiedergegeben wird, worin die Aminogruppen der Aminophenoxygruppen in Para-Stellung gebunden sind, reduziert wird.

## 7. Verfahren nach Anspruch 5, worin

- (i) die Verbindung mit einer Struktur der Formel (II) ausgewählt ist aus 1,4- und 1,3-Bis-(4-hydroxy- $\alpha,\alpha$ -dimethylbenzyl)-benzol, die Nitrobenzolverbindung 4-Chlornitrobenzol ist und worin 1,4- oder 1,3-Bis[4-(4-amino-phenoxy)- $\alpha,\alpha$ -dimethylbenzyl]-benzol erhalten wird; oder
- (ii) die Verbindung mit einer Struktur, die durch die Formel (II) repräsentiert ist, ausgewählt wird aus 1,4- und 1,3-Bis(4-hydroxy- $\alpha,\alpha$ -dimethylbenzyl)-benzol, die Nitrobenzolverbindung m-Dinitrobenzol ist und worin 1,4- oder 1,3-Bis[4-(3-aminophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]-benzol erhalten wird; oder
- (iii) die Verbindung mit einer Struktur, die durch die Formel (II) repräsentiert ist, ausgewählt wird aus 1,4- und 1,3-Bis-(4-hydroxy-3,5-dimethyl- $\alpha,\alpha$ -dimethylbenzyl)-benzol, die Nitrobenzolverbindung 4-Chlornitrobenzol ist und worin 1,4- oder 1,3-Bis[4-(4-aminophenoxy)-3,5-dimethyl- $\alpha,\alpha$ -dimethylbenzyl]-benzol erhalten wird; oder
- (iv) die Verbindung mit einer Struktur, die durch die Formel (II) repräsentiert wird, ausgewählt ist aus 1,4- und 1,3-Bis(4-hydroxy-3,5-dimethyl- $\alpha,\alpha$ -dimethylbenzyl)-benzol, die Nitrobenzolverbindung m-Dinitrobenzol ist und worin 1,4- oder 1,3-Bis[4-(3-aminophenoxy)-3,5-dimethyl- $\alpha,\alpha$ -dimethylbenzyl]-benzol erhalten wird.

8. Verfahren nach einem der Ansprüche 5 bis 7 unter Verwendung von zwei bis drei Molen der Nitrobenzolverbindung pro Mol der Verbindung mit einer Struktur, die durch die Formel (II) repräsentiert wird.

9. Verfahren nach einem der Ansprüche 5 bis 8, worin besagte Base wenigstens eine Verbindung ist, die ausgewählt ist aus Carbonaten, Hydrogencarbonaten, Hydroxiden und Alkoxiden der Alkalimetalle.

10. Verfahren nach einem der Ansprüche 5 bis 9, worin besagte Base in einer Menge von 2 Äquivalenten oder mehr verwendet wird, bezogen auf die Menge der besagten Verbindung mit einer Struktur, die durch die Formel (II) repräsentiert wird.

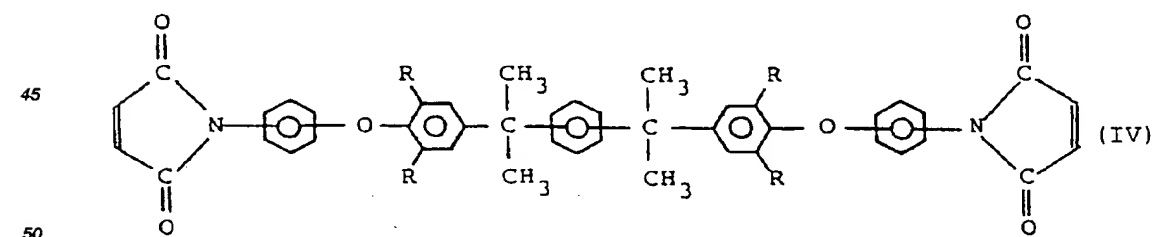
11. Verfahren nach einem der Ansprüche 5 bis 10, worin das Lösungsmittel wenigstens eines aus Dimethylsulfoxid, N,N-Dimethylformamid, N,N'-Dimethylacetamid, N-Methylpyrrolidon, 1,3-Dimethyl-2-imidazolidinon, Hexamethylphosphortriamid und Sulfolan ist.

12. Verfahren nach einem der Ansprüche 5 bis 11, worin das besagte Lösungsmittel in einer Menge verwendet wird, die auf das Gewicht bezogen das 1- bis 15-fache der Menge der Verbindung ist, mit einer Struktur, die durch die Formel (II) repräsentiert wird.

13. Verfahren nach einem der Ansprüche 5 bis 12, worin die Reaktion zwischen der Verbindung mit einer Struktur der Formel (II) und der Nitrobenzolverbindung im Bereich von 50 bis 220°C durchgeführt wird.

14. Verfahren nach einem der Ansprüche 5 bis 13, worin die Verbindung mit einer Struktur, die durch die Formel (III) repräsentiert wird, durch katalytische Reduktion oder Hydrazin-Reduktion reduziert wird.

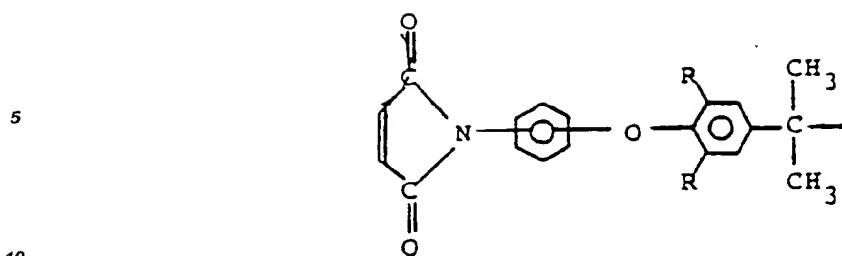
15. Bismaleimidverbindung mit einer Struktur, die durch die Formel (IV) repräsentiert wird



worin die R-Gruppen gleich oder unterschiedlich sind und H oder CH<sub>3</sub> darstellen.

16. Verbindung nach Anspruch 15, worin die Maleimidogruppen der Maleimidophenoxygruppen beide in Meta-Stellungen oder beide in Para-Stellungen gebunden sind.

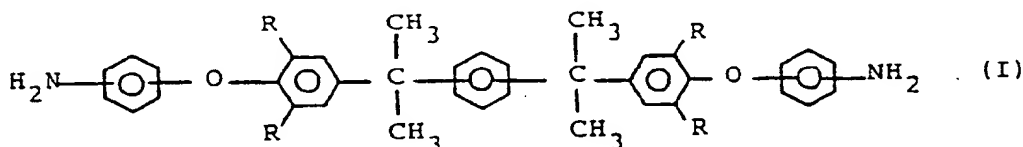
17. Verbindung nach Anspruch 15, worin die zwei durch die Formel



repräsentierten Gruppen beide in Meta-Stellungen oder beide in Para-Stellungen gebunden sind.

18. Verbindung nach Anspruch 15, ausgewählt aus 1,4- und 1,3-Bis[4-(4-maleimidophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]-benzol, 1,4- und 1,3-Bis[4-(3-maleimidophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]-benzol, 1,4- und 1,3-Bis[4-(4-maleimidophenoxy)-3,5-dimethyl- $\alpha,\alpha$ -dimethylbenzyl]-benzol und 1,4- und 1,3-Bis[4-(3-maleimidophenoxy)-3,5-dimethyl- $\alpha,\alpha$ -dimethylbenzyl]-benzol.

19. Verfahren zur Herstellung einer Verbindung nach einem der Ansprüche 15 bis 18, wobei das Verfahren die Reaktion einer aromatischen Diaminverbindung umfaßt, die durch die Formel (I)



wiedergegeben wird, worin die R-Gruppen gleich oder unterschiedlich sein können und H oder CH<sub>3</sub> darstellen, mit Maleinsäureanhydrid in Anwesenheit eines Säurekatalysators unter Erhitzungs/Rückfluß-Bedingungen eines organischen Lösungsmittels, wobei das erzeugte Wasser aus dem System entfernt wird.

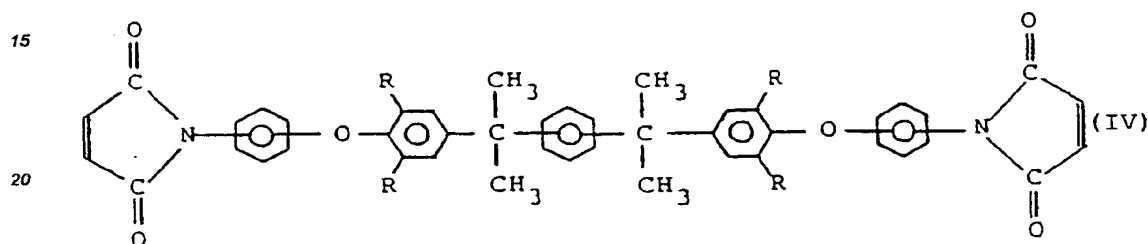
20. Verfahren nach Anspruch 19, worin die aromatische Diaminverbindung ausgewählt ist aus der Gruppe aus 1,4- und 1,3-Bis[4-(4-aminophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]-benzol, wobei 1,4- oder 1,3-Bis[4-(4-maleimidophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]-benzol erhalten wird; oder aus 1,4- und 1,3-Bis[4-(3-aminophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]benzol, wodurch 1,4- oder 1,3-Bis[4-(3-maleimidophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]-benzol erhalten wird; oder aus 1,4- und 1,3-Bis[4-(4-aminophenoxy)-3,5-dimethyl- $\alpha,\alpha$ -dimethylbenzyl]-benzol, wodurch 1,4- oder 1,3-Bis[4-(4-maleimidophenoxy)-3,5-dimethyl- $\alpha,\alpha$ -dimethylbenzyl]-benzol erhalten wird; oder aus 1,4- und 1,3-Bis[4-(3-aminophenoxy)-3,5-dimethyl- $\alpha,\alpha$ -dimethylbenzyl]-benzol, wodurch 1,4- oder 1,3-Bis[4-(3-maleimidophenoxy)-3,5-dimethyl- $\alpha,\alpha$ -dimethylbenzyl]-benzol erhalten wird.

21. Verfahren nach Anspruch 19 oder 20, worin der besagte Säurekatalysator wenigstens ein solcher ist aus Mineralsäuren, Heteropolysäuren, organischen Sulfonsäuren, halogenierten Carbonsäuren, Festsäuren, Diaminsalzen derselben und Kationenaustausch- und Ionenaustauschharzen; und der vorzugsweise wenigstens ein Glied ist, ausgewählt aus der Gruppe aus Schwefelsäure, Chlorwasserstoffsäure, Phosphorsäure, Phosphorwolframsäure, Phosphormolybdänsäure, p-Toluolsulfonsäure, Methansulfonsäure, Trichloressigsäure, Trifluoressigsäure, Silizium-Aluminiumoxid-Feststoffsäure und Diaminsalze derselben.

22. Verfahren nach Anspruch 19, 20 oder 21, worin der Säurekatalysator in einer Menge von 0,1 bis 10 Gew.-%, bezogen auf das Gesamtgewicht der aromatischen Diaminverbindung und des Maleinsäureanhydrids verwendet wird.

23. Verfahren nach einem der Ansprüche 19 bis 22, worin das organische Lösungsmittel wenigstens ein solches ist aus aliphatischen Kohlenwasserstoffen, alicyclischen Kohlenwasserstoffen, aromatischen Kohlenwasserstoffen, Halogeniden der aliphatischen und aromatischen Kohlenwasserstoffe und sauerstoffhaltigen, stickstoffhaltigen und schwefelhaltigen polaren Lösungsmitteln.

24. Verfahren nach einem der Ansprüche 19 bis 23, worin das organische Lösungsmittel in einer Menge verwendet wird, die das 1- bis 20-fache des Gesamtgewichtes des besagten aromatischen Diamins und des Maleinsäureanhydrids beträgt.
25. Verfahren nach einem der Ansprüche 19 bis 24, worin die Reaktion bei einer Temperatur im Bereich von 80 bis 190°C durchgeführt wird.
26. Verfahren nach einem der Ansprüche 19 bis 25, worin das molare Verhältnis von Maleinsäureanhydrid zur aromatischen Diaminverbindung 2,05 zu 3 beträgt.
27. Harzbildende Zusammensetzung, enthaltend die besagte aromatische Diaminverbindung nach einem der Ansprüche 1 bis 4 und eine Bismaleimidverbindung mit einer Struktur, die durch die Formel IV

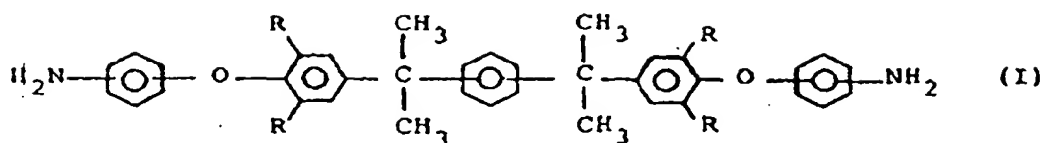


wiedergegeben wird, worin die R-Gruppen gleich oder unterschiedlich sein können und H oder CH<sub>3</sub> darstellen, die besagte aromatische Diaminverbindung in einem Verhältnis von 0,1 bis 1,2 Molen pro Mol der besagten Bis-Maleimidverbindung anwesend ist.

28. Harzbildende Zusammensetzung, enthaltend eine Bismaleimidverbindung nach einem der Ansprüche 15 bis 18 und eine aromatische Diaminverbindung, ausgewählt aus 1,4- und 1,3-Bis[4-(4-aminophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]-benzol, 1,4- und 1,3-Bis[4-(3-aminophenoxy)- $\alpha,\alpha$ -dimethylbenzyl]-benzol, 1,4- und 1,3-Bis[4-(4-aminophenoxy)-3,5-dimethyl- $\alpha,\alpha$ -dimethylbenzyl]-benzol und 1,4- und 1,3-Bis[4-(3-aminophenoxy)-3,5-dimethyl- $\alpha,\alpha$ -dimethylbenzyl]-benzol; wobei die aromatische Diaminverbindung in einem Verhältnis von 0,1 bis 1,2 Molen pro Mol der besagten Bismaleimidverbindung anwesend ist.
29. Harz der harzbildenden Zusammensetzung nach Anspruch 27 oder 28.
30. Harzbildende Zusammensetzung nach Anspruch 27 oder 28, die wenigstens ein Additiv enthält, welches ausgewählt ist aus einem Härtungsbeschleuniger, einem verstärkenden Material, einem Füllstoff und einem synthetischen Harz.
31. Verfahren zur Herstellung einer Harzzusammensetzung nach Anspruch 27 oder 28, umfassend (i) das Mischen der besagten Bismaleimidverbindung mit der besagten aromatischen Diaminverbindung und anschließend (ii) Unterwerfen der erhaltenen Mischung einer Hitzebehandlung bei einer Temperatur von 70 bis 220°C zur Durchführung der Prepolymerisation.
32. Verfahren nach Anspruch 31, bei dem die Bismaleimidverbindung und die aromatische Diaminverbindung in Anwesenheit eines organischen Lösungsmittels gemischt werden.

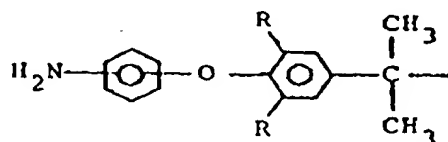
## Revendications

1. Une diamine aromatique ayant une structure représentée par la formule (I) :



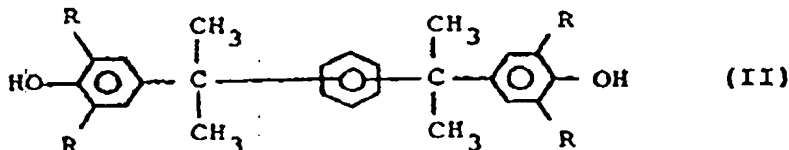
dans laquelle les groupes R peuvent être identiques ou différents et sont H ou CH<sub>3</sub>.

2. Un composé selon la revendication 1, où les groupes amino des groupes aminophénoxy sont tous deux liés dans les positions méta ou tous deux liés dans les positions para.
3. Un composé selon la revendication 1, où les deux groupes représentés par la formule :

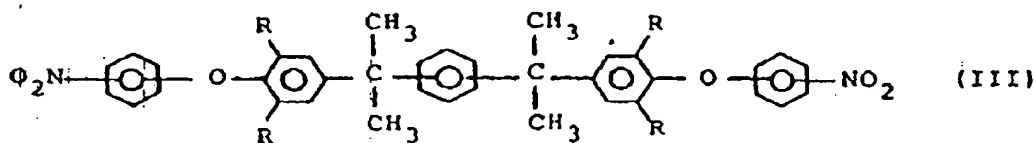


sont tous deux liés dans les positions méta ou tous deux liés dans les positions para.

4. Un composé selon la revendication 1 choisi parmi le 1,4- et le 1,3-bis[4-(4-aminophénoxy)- $\alpha,\alpha$ -diméthylbenzyl]benzène, le 1,4- et le 1,3-bis[4-(3-aminophénoxy)- $\alpha,\alpha$ -diméthylbenzyl]benzène, le 1,4- et le 1,3-bis[4-(4-aminophénoxy)-3,5-diméthyl- $\alpha,\alpha$ -diméthylbenzyl] benzène et le 1,4- et le 1,3-bis[4-(3-aminophénoxy)-3,5-diméthyl- $\alpha,\alpha$ -diméthylbenzyl]benzène.
5. Un procédé pour préparer un composé selon l'une quelconque des revendications précédentes, ledit procédé comprenant (i) la réaction d'un composé ayant une structure représentée par la formule (II) :



dans laquelle les groupes R peuvent être identiques ou différents et sont H ou CH<sub>3</sub>, avec un dérivé de nitrobenzène choisi parmi les nitrochlorobenzènes et les dinitrobenzènes, dans un solvant polaire aprotique, en présence d'une base, pour former un composé ayant une structure représentée par la formule (III) :



dans laquelle les groupes R peuvent être identiques ou différents et sont H ou CH<sub>3</sub>, puis (ii) la réduction du composé ayant une structure représentée par la formule (III).

6. Un procédé selon la revendication 5, dans lequel le dérivé de nitrobenzène est le m-dinitrobenzène et le composé ayant une structure représentée par la formule (III) est réduit pour produire un composé ayant une structure représentée par la formule (I) dans laquelle les groupes amino des groupes aminophénoxy sont liés dans les positions méta ,  
ou dans lequel  
le dérivé de nitrobenzène est le 4-chloronitrobenzène et le composé ayant une structure représentée par

la formule (III) est réduit pour produire un composé ayant une structure représentée par la formule (I) dans laquelle les groupes amino des groupes aminophénoxy sont liés dans les positions para.

7. Un procédé selon la revendication 5, dans lequel :

- (i) le composé ayant une structure de formule (II) est choisi parmi le 1,4- et le 1,3-bis(4-hydroxy- $\alpha,\alpha$ -diméthylbenzyl)benzène, le dérivé de nitrobenzène est le 4-chloronitrobenzène et on obtient le 1,4- ou le 1,3-bis[4-(4-aminophénoxy)- $\alpha,\alpha$ -diméthylbenzyl]benzène ; ou
- (ii) le composé ayant une structure représentée par la formule (II) est choisi parmi le 1,4- et le 1,3-bis(4-hydroxy- $\alpha,\alpha$ -diméthylbenzyl)benzène, le dérivé du nitrobenzène est le m-dinitrobenzène et on obtient le 1,4- ou le 1,3-bis[4-(3-aminophénoxy)- $\alpha,\alpha$ -diméthylbenzyl]benzène ; ou
- (iii) le composé ayant une structure représentée par la formule (II) est choisi parmi le 1,4- et le 1,3-bis(4-hydroxy-3,5-diméthyl- $\alpha,\alpha$ -diméthylbenzyl)benzène, le dérivé du nitrobenzène est le 4-chloronitrobenzène et on obtient le 1,4- ou le 1,3-bis[4-(4-aminophénoxy)-3,5-diméthyl- $\alpha,\alpha$ -diméthylbenzyl]benzène ; ou
- (iv) le composé ayant une structure représentée par la formule (II) est choisi parmi le 1,4- et le 1,3-bis(4-hydroxy-3,5-diméthyl- $\alpha,\alpha$ -diméthylbenzyl)benzène, le dérivé du nitrobenzène est le m-dinitrobenzène et on obtient le 1,4- ou le 1,3-bis[4-(3-aminophénoxy)-3,5-diméthyl- $\alpha,\alpha$ -diméthylbenzyl]benzène.

8. Le procédé selon l'une quelconque des revendications 5 à 7 utilisant 2 à 3 moles du dérivé de nitrobenzène par mole dudit composé ayant une structure représentée par la formule (II).

9. Un procédé selon l'une quelconque des revendications 5 à 8, dans lequel ladite base est au moins un composé choisi parmi les carbonates, hydrogénocarbonates, hydroxydes et alcoolates de métaux alcalins.

10. Un procédé selon l'une quelconque des revendications 5 à 9, dans lequel ladite base est utilisée en une quantité de 2 équivalents ou plus relativement à la quantité dudit composé ayant une structure représentée par la formule (II).

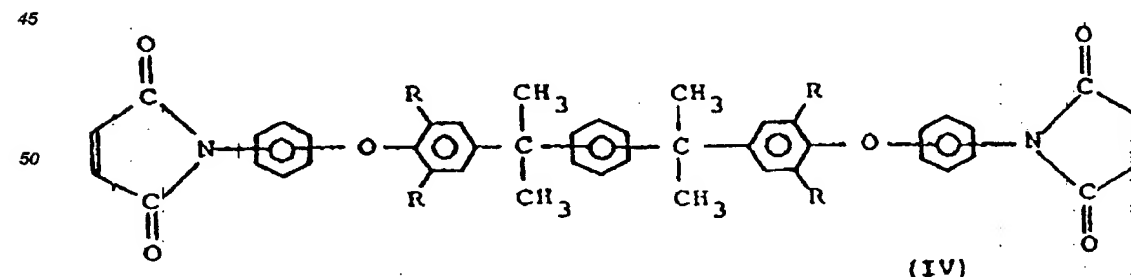
11. Un procédé selon l'une quelconque des revendications 5 à 10, dans lequel ledit solvant est au moins un du diméthylsulfoxyde, du N,N-diméthylformamide, du N,N'-diméthylacétamide, de la N-méthylpyrrolidone, de la 1,3-diméthyl-2-imidazolidinone, de l'hexaméthylphosphorotriamide et du sulfolane.

12. Un procédé selon l'une quelconque des revendications 5 à 11, dans lequel ledit solvant est utilisé en une quantité qui est de 1 à 15 fois en poids la quantité dudit composé ayant une structure représentée par la formule (II).

13. Un procédé selon l'une quelconque des revendications 5 à 12, dans lequel la réaction entre le composé ayant une structure de formule (II) et le dérivé de nitrobenzène est réalisée dans la gamme de 50 à 220°C.

14. Un procédé selon l'une quelconque des revendications 5 à 13, dans lequel le composé ayant une structure représentée par la formule (III) est réduit par réduction catalytique ou réduction par l'hydrazine.

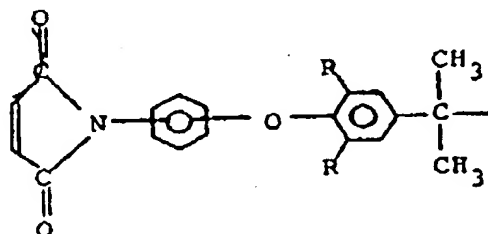
15. Un bismaléimide ayant une structure représentée par la formule (IV) :



dans laquelle les groupes R peuvent être identiques ou différents et sont H ou CH<sub>3</sub>.

16. Un composé selon la revendication 15, dans lequel les groupes maléimido des groupes maléimidophénoxy sont tous deux liés dans les positions méta ou tous deux liés dans les positions para.

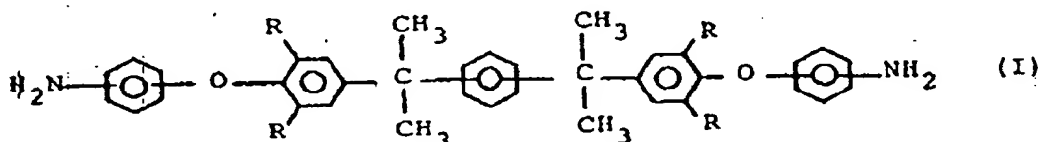
17. Un composé selon la revendication 15, dans lequel les deux groupes représentés par la formule :



sont tous deux liés dans les positions méta ou tous deux liés dans les positions para.

18. Le composé selon la revendication 15 choisi parmi le 1,4- et le 1,3-bis[4-(4-maleimidophénoxy)- $\alpha,\alpha$ -diméthylbenzyl]benzène, le 1,4- et le 1,3-bis[4-(3-maleimidophénoxy)- $\alpha,\alpha$ -diméthylbenzyl]benzène, le 1,4- et le 1,3-bis[4-(4-maleimidophénoxy)-3,5-diméthyl- $\alpha,\alpha$ -diméthylbenzyl]benzène et le 1,4- et le 1,3-bis[4-(3-maleimidophénoxy)-3,5-diméthyl- $\alpha,\alpha$ -diméthylbenzyl]benzène.

19. Un procédé pour préparer un composé selon l'une quelconque des revendications 15 à 18, ledit procédé comprenant la réaction d'une diamine aromatique représentée par la formule (I) :



dans laquelle les groupes R peuvent être identiques ou différents et sont H ou CH<sub>3</sub>, avec l'anhydride maléique, en présence d'un catalyseur acide avec chauffage/reflux d'un solvant organique, et l'élimination hors du système de toute eau produite.

20. Un procédé selon la revendication 19, dans lequel la diamine aromatique est choisie dans le groupe constitué du 1,4- et du 1,3-bis[4-(4-aminophénoxy)- $\alpha,\alpha$ -diméthylbenzyl]benzène, pour obtenir le 1,4- ou le 1,3-bis[4-(4-maleimidophénoxy)- $\alpha,\alpha$ -diméthylbenzyl]benzène ;  
ou parmi le 1,4- et le 1,3-bis[4-(3-aminophénoxy)- $\alpha,\alpha$ -diméthylbenzyl]benzène, pour obtenir le 1,4- ou le 1,3-bis[4-(3-maleimidophénoxy)- $\alpha,\alpha$ -diméthylbenzyl]benzène ;  
ou parmi le 1,4- et le 1,3-bis[4-(4-aminophénoxy)-3,5-diméthyl- $\alpha,\alpha$ -diméthylbenzyl]benzène, pour obtenir le 1,4- ou le 1,3-bis[4-(4-maleimidophénoxy)-3,5-diméthyl- $\alpha,\alpha$ -diméthylbenzyl]benzène ;  
ou parmi le 1,4- et le 1,3-bis[4-(3-aminophénoxy)-3,5-diméthyl- $\alpha,\alpha$ -diméthylbenzyl]benzène, pour obtenir le 1,4- ou le 1,3-bis[4-(3-maleimidophénoxy)-3,5-diméthyl- $\alpha,\alpha$ -diméthylbenzyl]benzène.

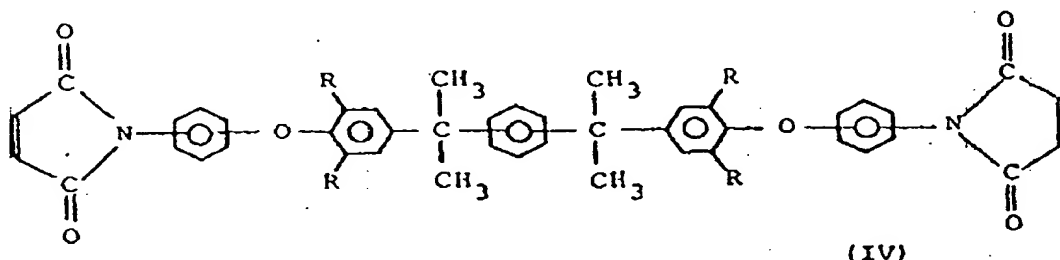
21. Un procédé selon la revendication 19 ou 20, dans lequel ledit catalyseur acide est au moins un des acides minéraux, hétéropolyacides, acides organiques sulfoniques, acides carboxyliques alogénés, acides solides, leurs sels de diamine et résines échangeuses d'ions de type échangeuses de cations, et est de préférence au moins un composant de l'ensemble constitué par l'acide sulfurique, l'acide chlorhydrique, l'acide phosphorique, l'acide phosphotungstique, l'acide phosphomolybdique, l'acide p-toluènesulfonique, l'acide méthanesulfonique, l'acide trichloroacétique, l'acide trifluoroacétique, un acide solide de type silicone-alumine et leurs sels de diamine.

22. Un procédé selon la revendication 19, 20 ou 21, dans lequel le catalyseur acide est utilisé en une quantité de 0,1 à 10 % en poids relativement au poids total de la diamine aromatique et de l'anhydride maléique.

23. Un procédé selon l'une quelconque des revendications 19 à 22, dans lequel le solvant organique est au moins un des hydrocarbures aliphatiques, hydrocarbures alicycliques, hydrocarbures aromatiques, halo-

génères desdits hydrocarbures aliphatiques et aromatiques et solvants polaires contenant de l'oxygène, contenant de l'azote et contenant du soufre.

24. Un procédé selon l'une quelconque des revendications 19 à 23, dans lequel ledit solvant organique est utilisé en une quantité de 1 à 20 fois le poids total de ladite diamine aromatique et de l'anhydride maléique.
25. Le procédé selon l'une quelconque des revendications 19 à 24, dans lequel la réaction est effectuée à une température dans la gamme de 80 à 190°C.
26. Un procédé selon l'une quelconque des revendications 19 à 25, dans lequel le rapport molaire de l'anhydride maléique à la diamine aromatique est de 2,05 à 3.
27. Une composition génératrice de résine comprenant ladite diamine aromatique selon l'une quelconque des revendications 1 à 4 et un bismaléimide ayant une structure représentée par la formule (IV) :



dans laquelle les groupes R peuvent être identiques ou différents et sont H ou CH<sub>3</sub>, ladite diamine aromatique étant présente dans un rapport de 0,1 à 1,2 mole par mole dudit bismaléimide.

28. Une composition génératrice de résine comprenant un bismaléimide selon l'une quelconque des revendications 15 à 18 et une diamine aromatique choisie parmi le 1,4- et le 1,3-bis[4-(4-aminophénoxy)- $\alpha,\alpha$ -diméthylbenzyl]benzène, le 1,4- et le 1,3-bis[4-(3-aminophénoxy)- $\alpha,\alpha$ -diméthylbenzyl]benzène, le 1,4- et le 1,3-bis[4-(4-aminophénoxy)-3,5-diméthyl- $\alpha,\alpha$ -diméthylbenzyl]benzène, et le 1,4- et le 1,3-bis[4-(3-aminophénoxy)-3,5-diméthyl- $\alpha,\alpha$ -diméthylbenzyl]benzène ; ladite diamine aromatique étant présente dans un rapport de 0,1 à 1,2 mole par mole dudit bismaléimide.
29. Une résine de la composition génératrice de résine selon la revendication 27 ou 28.
30. Une composition génératrice de résine selon la revendication 27 ou 28 comprenant de plus au moins un additif choisi parmi un accélérateur de durcissement, une matière de renfort, une charge et une résine synthétique.
31. Un procédé pour préparer une composition de résine selon la revendication 27 ou 28 comprenant (i) le mélange dudit bismaléimide avec ladite diamine aromatique, puis (ii) le traitement thermique du mélange à une température de 70 à 220°C pour effectuer une prépolymérisation.
32. Un procédé selon la revendication 31, dans lequel le bismaléimide et la diamine aromatique sont mélangés en présence d'un solvant organique.



FIG. 1

